







Combinatorics and Arithmetic for Physics

Wave fronts with cross-diffusion

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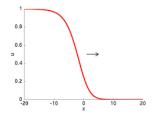






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with u(x,t) and a parameter rBelongs to the class of reaction-diffusion equations Admits two equilibrium states u = 0 and u = 1Existence of a travelling wave between the two states





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When $D_A = D_B$, the RD equation of A(x,t) is a Fisher-KPP equation.









$$\partial_t A = D_A \partial_x \left[\left(1 - \frac{A}{C} \right) \partial_x A \right] - D_B \partial_x \left(\frac{A}{C} \partial_x B \right) + kAB$$
$$\partial_t B = D_B \partial_x \left[\left(1 - \frac{B}{C} \right) \partial_x B \right] - D_A \partial_x \left(\frac{B}{C} \partial_x A \right) - kAB$$

where C = A + B + S =constant









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- How does $D_A \neq D_B$ affect the velocity and shape of the wave front ?
- How does confinement (i.e. $S \rightarrow 0$) affect the velocity and shape of the wave front ?
- Can we use these effects to detect perturbed diffusion in a concentrated Fisher-KPP front ?

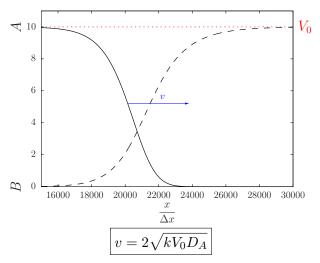








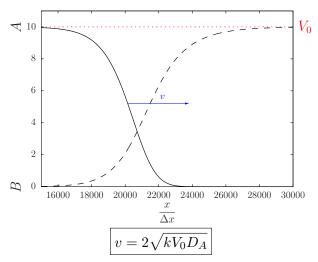
Velocity \boldsymbol{v} of a Fisher front







Velocity v of a Fisher front Independent of D_B and S













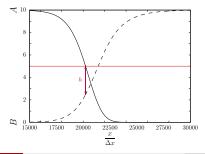






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$$A(x_0)$$
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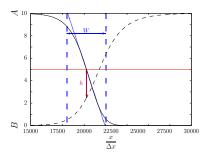








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 - $A(x_0)$ and $B(x_0)$ where $A(x_0) = \frac{V_0}{2}$
- Width of the front defined as the inverse of the slope of the A profile where $A(x_0)=\frac{V_0}{2}$

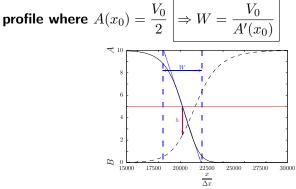








- Height defined as the difference of concentration between
 - $A(x_0)$ and $B(x_0)$ where $A(x_0) = \frac{V_0}{2} \Rightarrow h = B(x_0) \frac{V_0}{2}$
- \bullet Width of the front defined as the inverse of the slope of the A









To study the RD equations, we define the moving frame:

$$(x,t) \rightarrow (\zeta)$$

$$\zeta = \frac{x}{v} - t$$

$$\zeta_0 = \frac{x_0}{v} - t_0 \equiv 0$$

$$A(x,t) \rightarrow f(\zeta)$$

$$B(x,t) \rightarrow g(\zeta)$$

$$\epsilon = 1/v^2$$

$$d = \text{dilute (regular diffusion)}$$

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Examples:

$f_d = \mathbf{A}$ concentration in the dilute case

 $g_c = \mathbf{B}$ concentration in the concentrated case









Dilute case RD equations in the moving frame:

$$0 = kf_dg_d + f'_d + \epsilon D_A f''_d$$
$$0 = -kf_dg_d + g'_d + \epsilon D_B g''_d$$









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Concentrated case RD equations in the moving frame:

$$0 = kf_cg_c + f'_c + \epsilon \left(D_A \left[\left(1 - \frac{f'_c}{C} \right) f''_c - \frac{(f'_c)^2}{C} \right] - D_B \left[\frac{f_cg''_c}{C} + \frac{f'_cg'_c}{C} \right] \right) \\ 0 = -kf_cg_c + g'_c + \epsilon \left(D_B \left[\left(1 - \frac{g'_c}{C} \right) g''_c - \frac{(g'_c)^2}{C} \right] - D_A \left[\frac{g_cf''_c}{C} + \frac{f'_cg'_c}{C} \right] \right)$$







We consider k, V_0 and D_A such as:

 $\epsilon \ll 1$

Diffusion can be considered as a perturbation of reaction (do not mix up with perturbed diffusion !). We write f and q as a perturbation series:

$$f = f_0 + \epsilon f_1 + \epsilon^2 f_2 + \dots$$
$$g = g_0 + \epsilon g_1 + \epsilon^2 g_2 + \dots$$







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Zero-th order solutions (no diffusion) are straightforwardly obtained:

$$f_{d,0} = f_{c,0} = \frac{V_0}{1 + e^{kV_0\zeta}}$$
$$g_{d,0} = g_{c,0} = \frac{V_0}{1 + e^{-kV_0\zeta}}$$





Now, instead of deriving the solutions for the higher-order terms $f_{1,2,...}$ and $g_{1,2,...}$, we focus on the point $\zeta_0 = 0$.









Now, instead of deriving the solutions for the higher-order terms $f_{1,2,\ldots}$ and $g_{1,2,\ldots}$, we focus on the point $\zeta_0 = 0$. The expressions for the height and the width, up to the second-order, are given by:

$$\begin{split} h_d &= \frac{V_0}{16} \left(1 - \frac{D_B}{D_A} \right) \left[1 + \frac{1}{8} \left(1 - \frac{D_B}{D_A} \right) \right] \\ h_c &= \frac{V_0}{16} \left(1 - \frac{D_B}{D_A} \right) \left(1 - \frac{V_0}{C} \right) \left[1 + \frac{1}{8} \left(1 - \frac{D_B}{D_A} \right) \left(1 - 2\frac{V_0}{C} \right) \right] \\ W_d &= 8\sqrt{\frac{D_A}{kV_0}} \left[1 + \frac{1}{8} \left(1 - \frac{D_B}{D_A} \right) - \frac{1}{64} \frac{D_B}{D_A} \left(3 - \frac{D_B}{D_A} \right) \right]^{-1} \\ W_c &= 8\sqrt{\frac{D_A}{kV_0}} \left[1 + \frac{1}{8} \left(1 - \frac{D_B}{D_A} \right) \left(1 - \frac{3V_0}{2C} \right) - \frac{1}{64} \left[\frac{D_B}{D_A} \left(3 - \frac{D_B}{D_A} \right) \right]^{-1} \\ &+ \left(\frac{9}{2} - 8\frac{D_B}{D_A} + \frac{7}{2} \frac{D_B^2}{D_A^2} \right) \frac{V_0}{C} - \left(\frac{7}{2} - 7\frac{D_B}{D_A} + \frac{7}{2} \frac{D_B^2}{D_A^2} \right) \frac{V_0^2}{C^2} \right] \right]^{-1} \end{split}$$







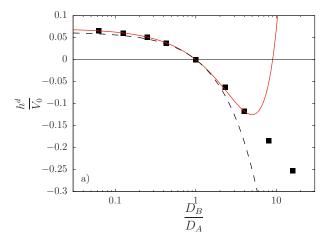


Comparison between analytic and numerical results:

G. Morgado (IPC, PAS/ LPTMC, SU)



Comparison between analytic and numerical results: Height in the diluted case

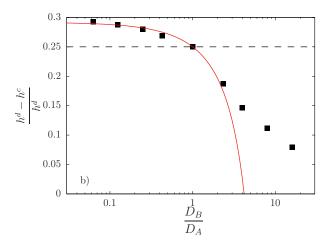




Perturbed diffusion ($V_0/C = 0.25$) vs. regular diffusion ($V_0/C = 0$):



Perturbed diffusion ($V_0/C = 0.25$) vs. regular diffusion ($V_0/C = 0$): Height









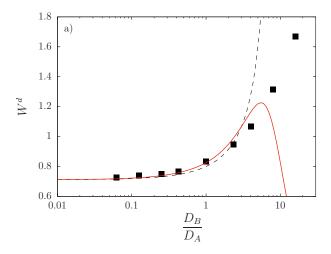


Comparison between analytic and numerical results:

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Comparison between analytic and numerical results: Width in the diluted case

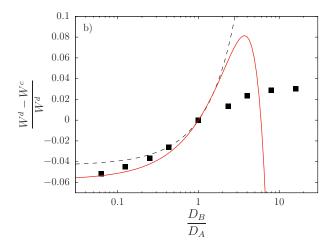




Perturbed diffusion ($V_0/C = 0.25$) vs. regular diffusion ($V_0/C \rightarrow 0$):



Perturbed diffusion ($V_0/C = 0.25$) vs. regular diffusion ($V_0/C \rightarrow 0$): Width











Concentration effects :

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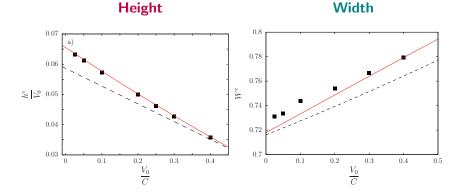








Concentration effects :



for $D_B/D_A = 1/16$









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- The shape of the front is affected by both effects. The height is strongly influenced by the ratio of diffusion coefficients D_B/D_A and the deviation from ideal solution V_0/C . However, its discriminating property is more efficient for low values of D_B/D_A .









- The velocity of the front is not affected by the diffusion of B species nor the perturbation of diffusion by solvent concentration.
- The shape of the front is affected by both effects. The height is strongly influenced by the ratio of diffusion coefficients D_B/D_A and the deviation from ideal solution V_0/C . However, its discriminating property is more efficient for low values of D_B/D_A .
- The width is also mainly affected by the ratio D_B/D_A and V_0/C , working better for large values of D_B/D_A .









Thank You

[1] Morgado, Nowakowski, Lemarchand, Phys. Rev. E 99, 022205 (2019)



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Cross-diffusion terms derivation from linear irreversible thermodynamics

The entropy production per unit mass due to isothermal diffusion is given by:

$$\sigma = \frac{1}{T} \sum_{X=A,B,S} \vec{j}_X \cdot \left(-\vec{\nabla}\mu_X\right)$$

where T is the temperature, j_X the flux of species X, and μ_X is the chemical potential of species X.

We consider the framework of the solvent. The flux of species X in this framework is defined by:

$$\begin{aligned} j_X^{\vec{*}} &= \rho_X \left(\vec{u}_X - \vec{u}_S \right) = \rho_X \left(\vec{u}_X - \vec{u} \right) + \rho_X \left(\vec{u} - \vec{u}_S \right) \\ &= \vec{j}_X - \frac{\rho_X}{\rho_S} \vec{j}_S \end{aligned}$$

where ρ_X is the concentration of X, and \vec{u}_X is the velocity of X.



Prigogine's theorem: Entropy production does not depend on the chosen framework. Therefore:

$$\sigma = \frac{1}{T} \sum_{X=A,B,S} \vec{j}_X \cdot \left(-\vec{\nabla}\mu_X\right) = \frac{1}{T} \sum_{X=A,B} j_X^{\vec{*}} \cdot \left(-\vec{\nabla}\mu_X\right)$$

Assuming that the solution is ideal, we can write:

$$\mu_X = \mu_X^0 + RT \ln \frac{\rho_X}{\rho}$$

Using the expression for $j_X^{\vec{*}}$ and μ_X , we get:

$$\begin{pmatrix} \vec{j}_A \\ \vec{j}_B \end{pmatrix} = \begin{pmatrix} 1 - \frac{\rho_A}{\rho} & -\frac{\rho_A}{\rho} \\ -\frac{\rho_B}{\rho} & 1 - \frac{\rho_B}{\rho} \end{pmatrix} \begin{pmatrix} \vec{j}_A^* \\ \vec{j}_B^* \end{pmatrix}$$



$$\begin{pmatrix} \vec{j}_A \\ \vec{j}_B \end{pmatrix} = - \begin{pmatrix} 1 - \frac{\rho_A}{\rho} & \frac{\rho_A}{\rho} \\ \frac{\rho_B}{\rho} & 1 - \frac{\rho_B}{\rho} \end{pmatrix} \begin{pmatrix} D_A \vec{\nabla} \rho_A \\ D_B \vec{\nabla} \rho_B \end{pmatrix}$$

Hypothesis:
$$j_X^* = -D_X \vec{\nabla} \rho_X$$
 (Fick's first law in solvent framework)
Then:

 $\begin{pmatrix} \vec{j}_A \\ \vec{j}_B \end{pmatrix} = \begin{pmatrix} 1 - \frac{\rho_A}{\rho} & -\frac{\rho_A}{\rho} \\ -\frac{\rho_B}{\rho} & 1 - \frac{\rho_B}{\rho} \end{pmatrix} \begin{pmatrix} \vec{j}_A^* \\ \vec{j}_B^* \end{pmatrix}$

NaMeS 🕅 Using the expression for j_X^* and μ_X , we get:

