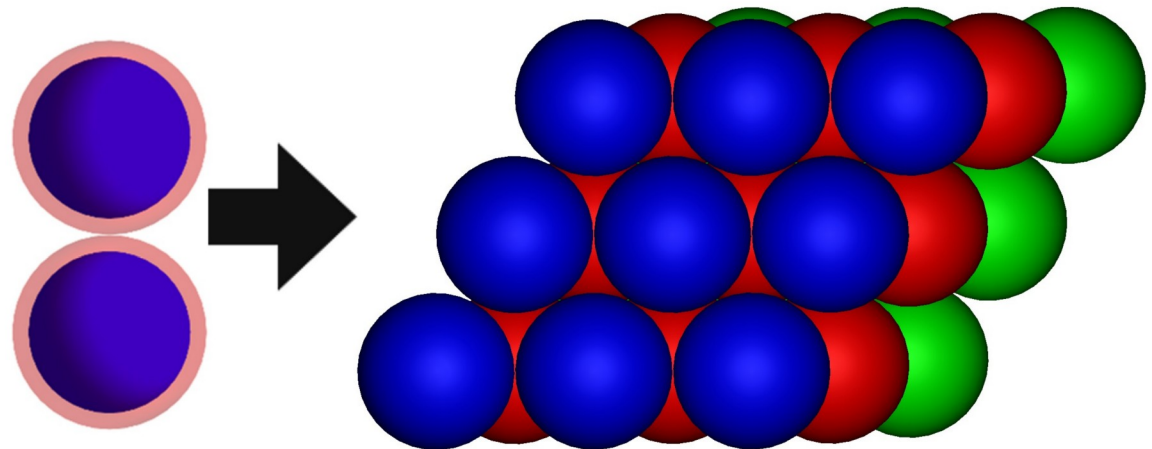


# Predicting self-assembly in simple models with multiple length scales

---

**Frank Smallenburg**  
Laboratoire de Physique des Solides  
Orsay



# ANR SoftQC



- Website : <https://softqc.wordpress.com/>

## Tackling quasicrystal self-assembly

- Nanoparticle synthesis
- Self-assembly in 2D and 3D
- Structural studies
  - interactions in solution :  $S(q)$
  - superlattices : Bragg peaks
- Simulations
- Theory

*Benjamin Abécassis (ENS Lyon)*

*Claire Goldmann*

*Brigitte Pansu*

*Marianne Impéror-Clerc*

*Giuseppe Foffi*

*Frank Smallenburg*

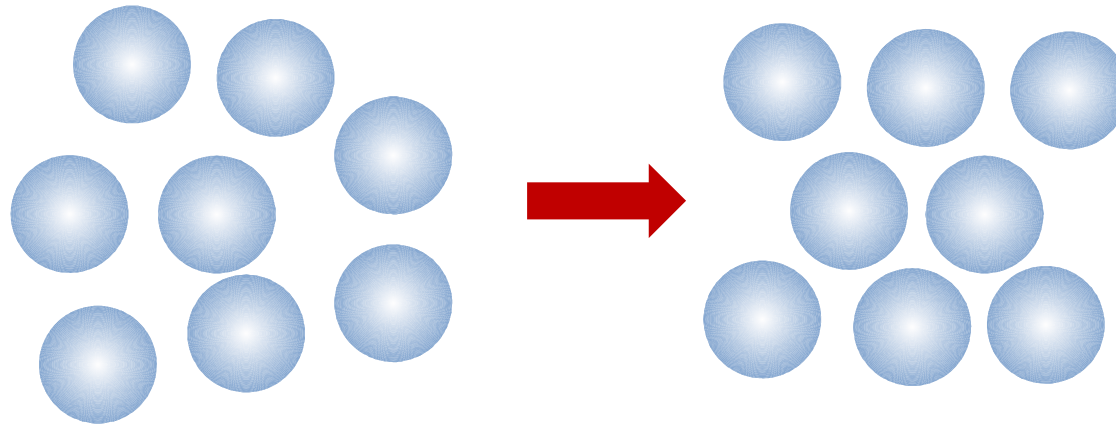
*Anuradha Jagannathan*

*Jean-François Sadoc*

2 PhD students (to start in October 2019) and 1 postdoc (to start in 2020)

# How can we predict self-assembly?

---



We have a set of particles with known interactions and conditions:

**What phase(s) can we expect to show up?**

# Predicting self-assembly with simulations

---



Direct strategy:

**Simulate the system, see what forms**

- Usually slow
- Not reliable
- Qualitative results

# Predicting self-assembly with simulations

---



Direct strategy:

**Simulate the system, see what forms**

- Usually slow
- Not reliable
- Qualitative results

**Can we be more precise?**

# Predicting phase behavior

---

Given:

- Number of particles **N**
- Volume **V**
- Temperature **T**

the thermodynamically stable state for any system is the state with the **lowest Helmholtz free energy**  $F(N, V, T)$ .

This free energy is a combination of potential energy and entropy:

$$F = U - TS$$

For large systems,  $F$  is proportional to  $N$  and  $V$ .

Hence, can be written as  $F(\rho, T) = N f(\rho, T)$ .

$$\rho = N / V$$

Number density

# Starting simple: Hard spheres

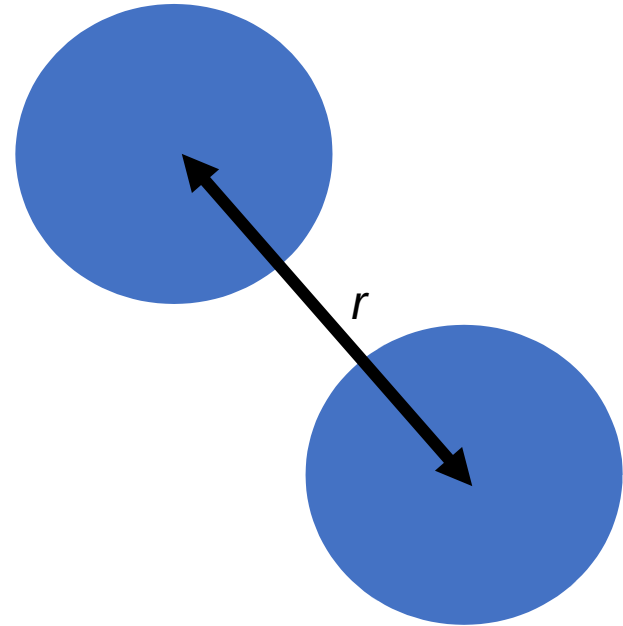
---

- **Model:**

$$\beta\phi(r) = \begin{cases} \infty, & r < \sigma \quad (\text{overlap}) \\ 0, & r \geq \sigma \quad (\text{no overlap}) \end{cases}$$

- **Helmholtz Free Energy:**

$$F = \cancel{U} - TS$$



For hard particles, all non-overlapping configurations have the same potential energy ( $U = 0$ ).

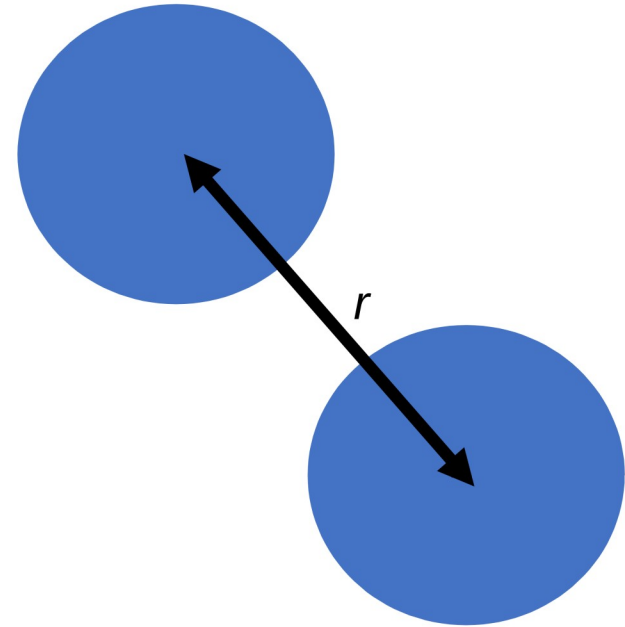
The **entropy** drives the phase transitions.

# Starting simple: Hard spheres

---

- **Model:**

$$\beta\phi(r) = \begin{cases} \infty, & r < \sigma \quad (\text{overlap}) \\ 0, & r \geq \sigma \quad (\text{no overlap}) \end{cases}$$



- **Helmholtz Free Energy:**

$$F = \cancel{U} - TS$$

$$F = -k_B T \log Q$$

$$Q(N, V, T) = \frac{1}{N! \Lambda^{3N}} \int d\mathbf{r}^N e^{-\beta U(\mathbf{r}^N)}$$

Boltzmann distribution

Thermal wavelength

$$\Lambda = \frac{h}{\sqrt{2\pi m k_B T}}$$

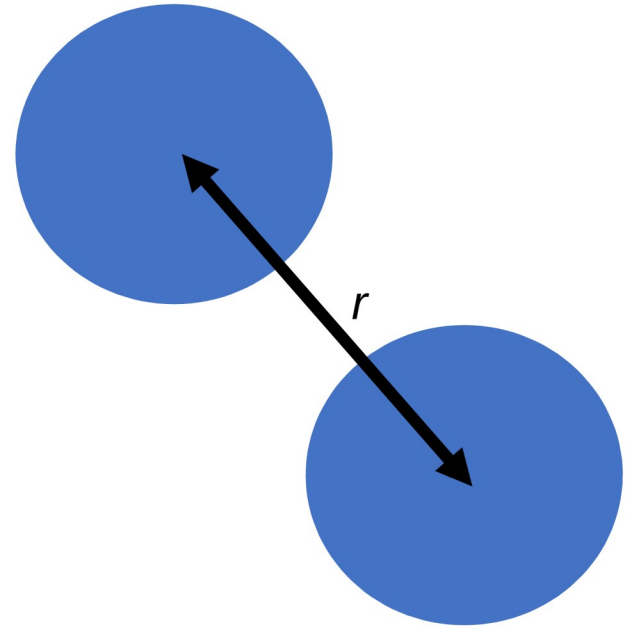


# Starting simple: Hard spheres

---

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Boltzmann distribution

No overlaps: 1  
Overlaps: 0

# Thermodynamic integration

---

Unfortunately, we **cannot** measure this entropy in simulations directly.

However, we can measure **derivatives** of the free energy.

For example the pressure:

The diagram illustrates the relationship between free energy, volume, and pressure. It features the following elements:

- A green box labeled "Volume" with an arrow pointing to the  $V$  in the denominator of the derivative  $\left(\frac{\partial F}{\partial V}\right)$ .
- The derivative  $\left(\frac{\partial F}{\partial V}\right)$  is shown in black, with the  $V$  in the denominator highlighted in green.
- A pink box labeled "N, T" is positioned below the derivative, with an arrow pointing to the derivative.
- An equals sign "=" is followed by a blue box containing the expression  $-p$ .
- A blue box labeled "Pressure" is positioned to the right of the  $-p$  box, with an arrow pointing from the  $-p$  box to it.
- A large pink box at the bottom right contains the text "Derivative taken at constant number of particles and temperature", with an arrow pointing from the "N, T" box to it.

$$\left(\frac{\partial F}{\partial V}\right)_{N,T} = -p$$

# Thermodynamic integration

---

Unfortunately, we **cannot** measure this entropy in simulations directly.

However, we can measure **derivatives** of the free energy.

For example the pressure:

The diagram illustrates the relationship between the derivative of free energy and pressure. It features the following elements:

- A central equation:  $\left(\frac{\partial F}{\partial V}\right)_{N,T} = -p$
- A green box labeled "Volume" with an arrow pointing to the  $V$  in the denominator of the derivative.
- A pink box labeled "N,T" with an arrow pointing to the subscript of the derivative.
- A blue box labeled "Pressure" with an arrow pointing to the  $p$  in the equation.
- A larger pink box containing the text "Derivative taken at constant number of particles and temperature" with an arrow pointing to the  $N,T$  subscript.

Additionally, in the low-density limit, the system behaves as an ideal gas, with a known free energy:

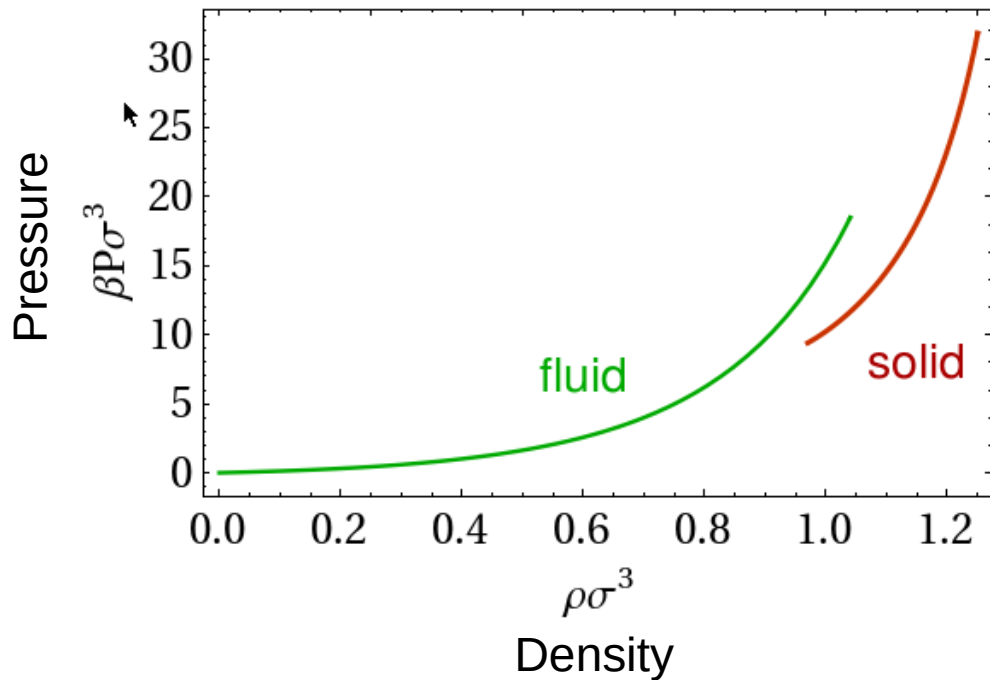
$$\frac{F_{id}}{Nk_B T} = \log(\rho\Lambda^3) - 1$$

Hence, we can calculate  $F$  at any density by integrating the pressure... **for a fluid**

# Hard sphere pressure

---

Measure **equation of state** in simulations:



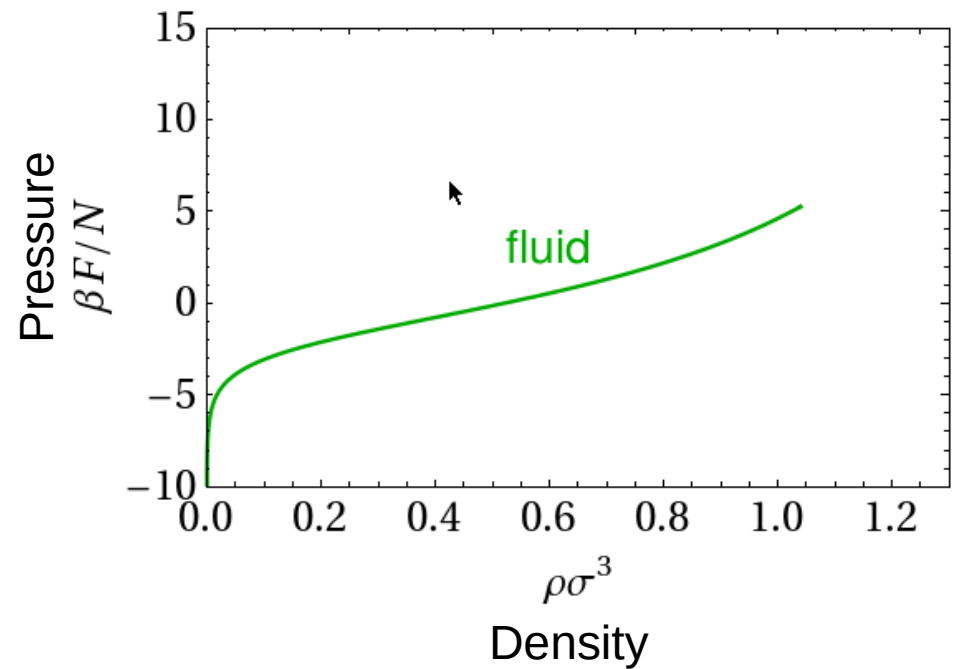
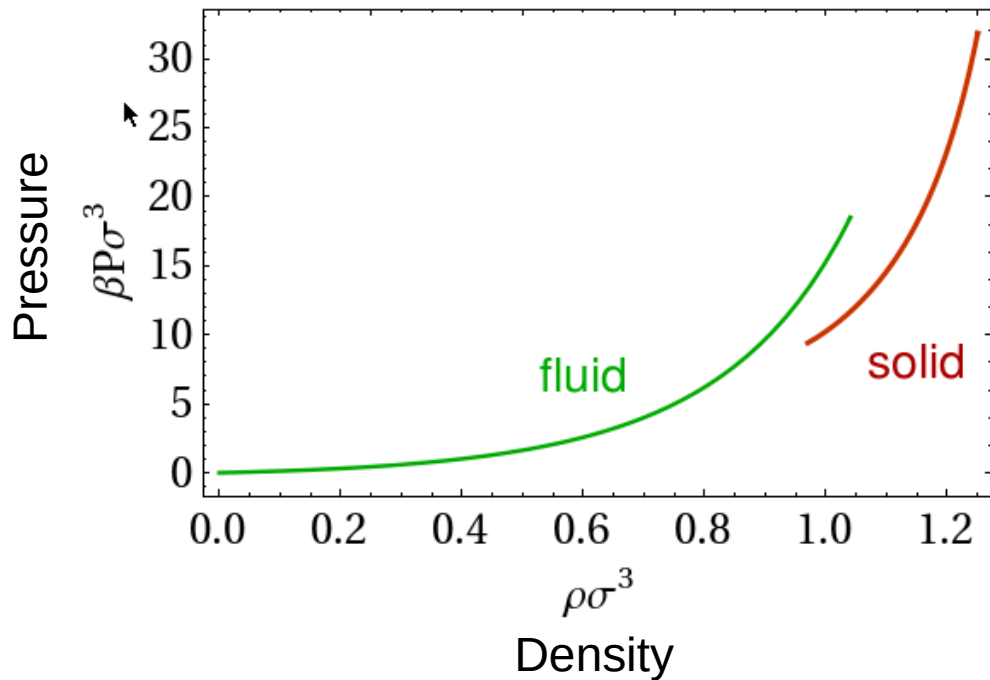
$$\rho = N/V$$

$$\beta = 1/k_B T$$

$\sigma$  = particle diameter

# Hard sphere pressure

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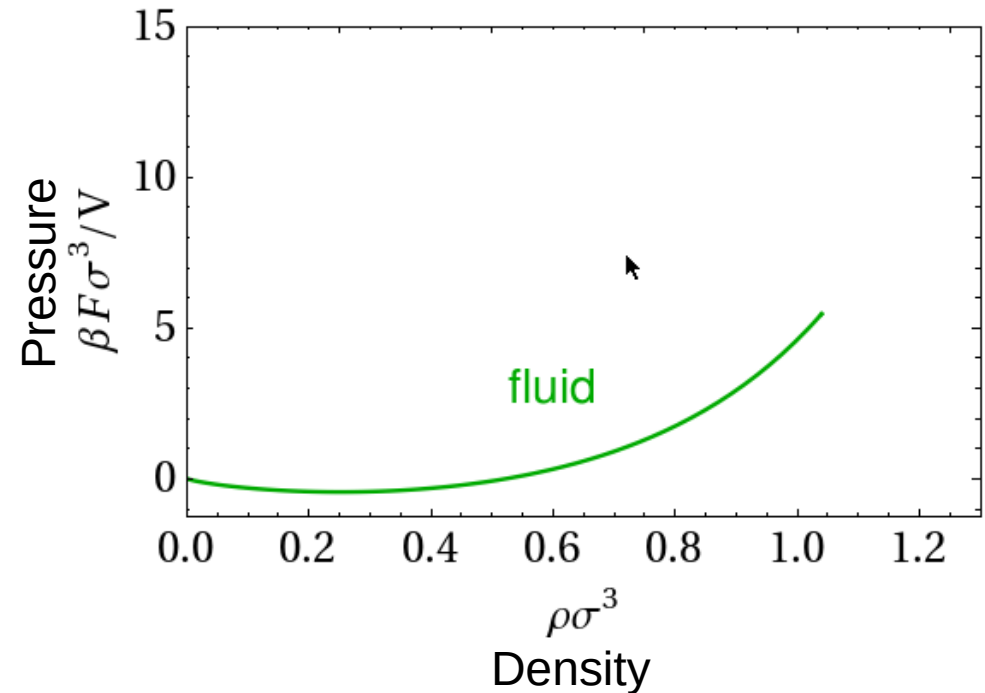
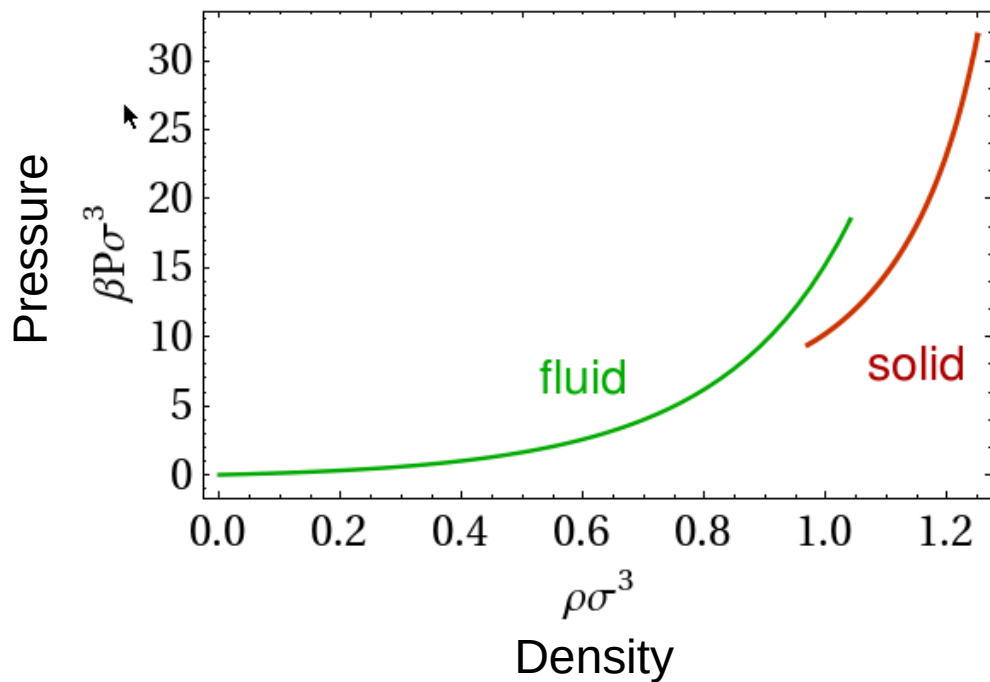
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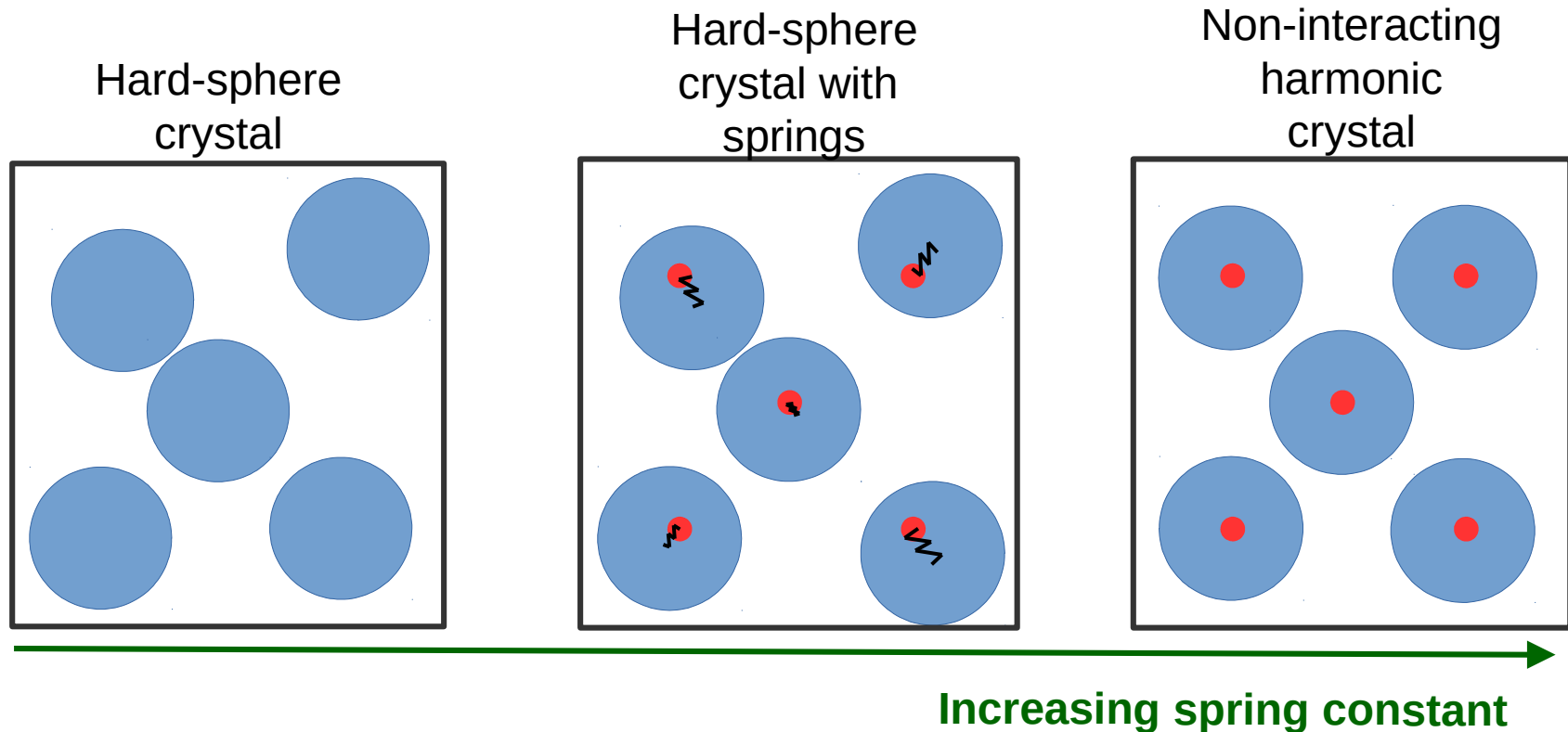
**But where do we draw the crystal?**

# Crystal free energy

For ordered phases, we need a **reference free energy**.

Several routes for obtaining a reference free energy:

## 1. Thermodynamic integration to an analytically solvable system



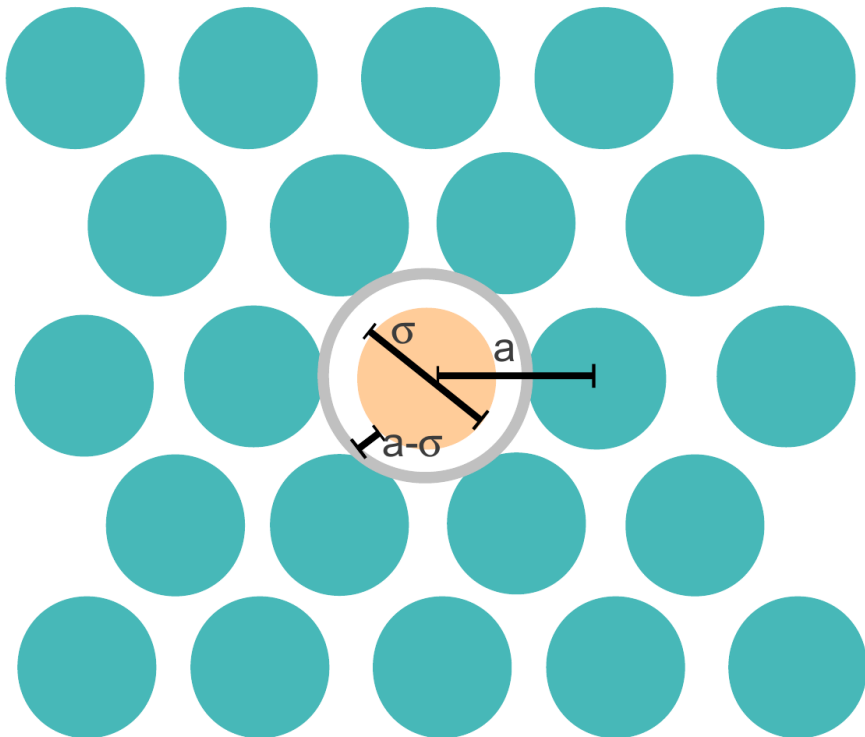
# Crystal free energy

---

For the crystal phase, we can do the same thing, if we know the free energy at some density.

Several routes for obtaining a reference free energy:

## 2. Theoretical approximations



Simple example: **Cell theory**

- Mean-field approximation of crystal free energy.
- Simple and reasonably accurate for hard particles.



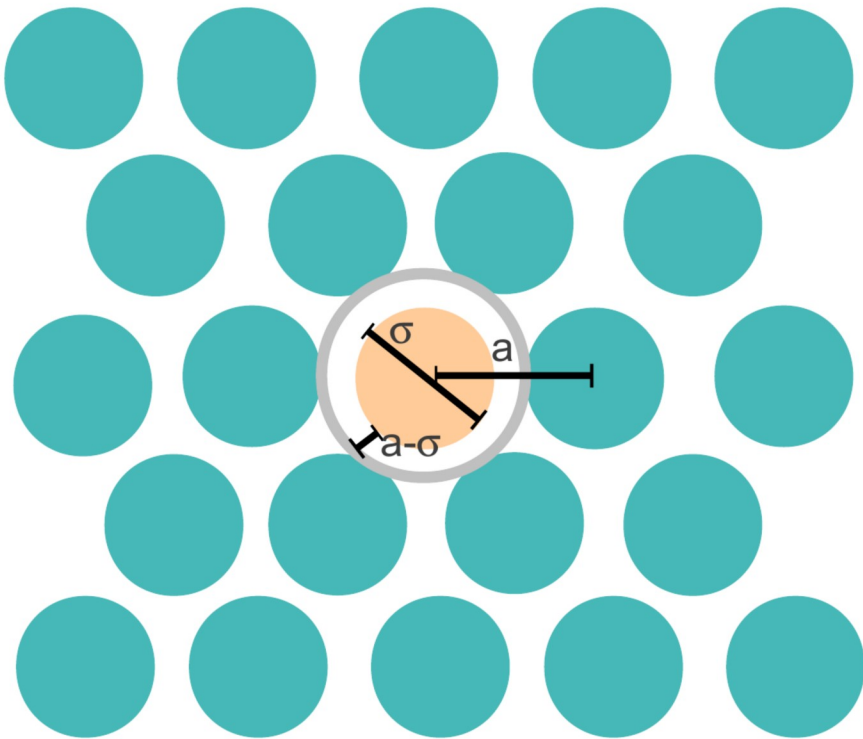
# Cell theory

Consider a hard-sphere crystal (e.g. FCC):

- lattice spacing:  $a = \left( \frac{\rho}{\rho_{\max}} \right)^{1/3}$

$$\rho = N / V$$

$$\rho_{\max} = \sqrt{2} / \sigma^3$$



If we assume all **other particles** are at their average positions, what is the free energy of a single particle?

$$\begin{aligned} f_1 &= -k_B T \log Z \\ &= -k_B T \log \left( \frac{1}{\Lambda^3} \int_{V_1} \mathbf{dr} \exp(-\beta U) \right) \\ &\simeq -k_B T \log \left( \frac{4\pi(a - \sigma)^3}{3} \right) \end{aligned}$$

Full free energy:

$$F = N f_1$$

# Cell theory

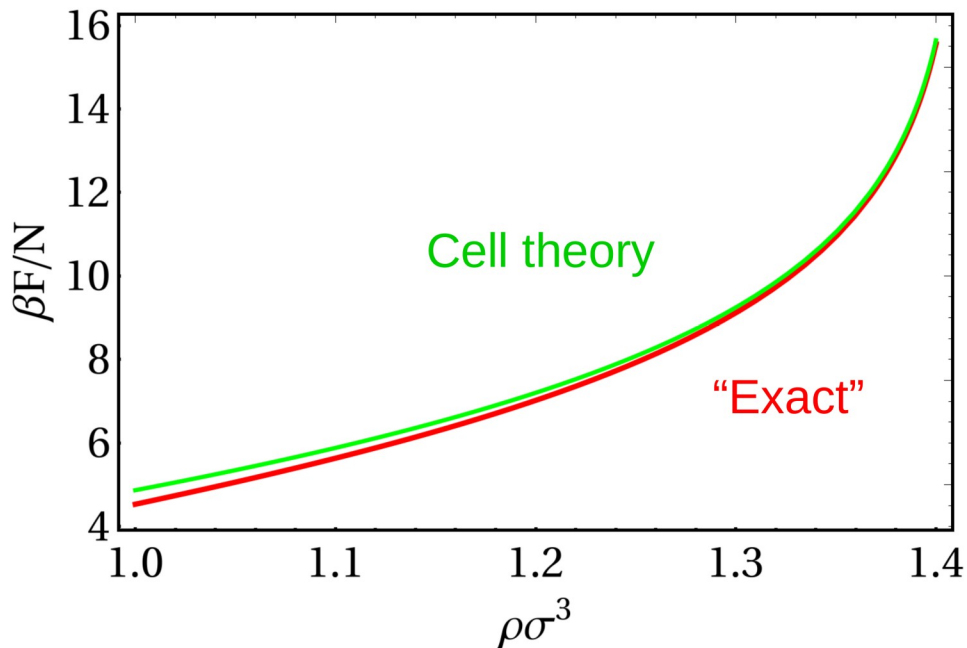
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# Cell theory

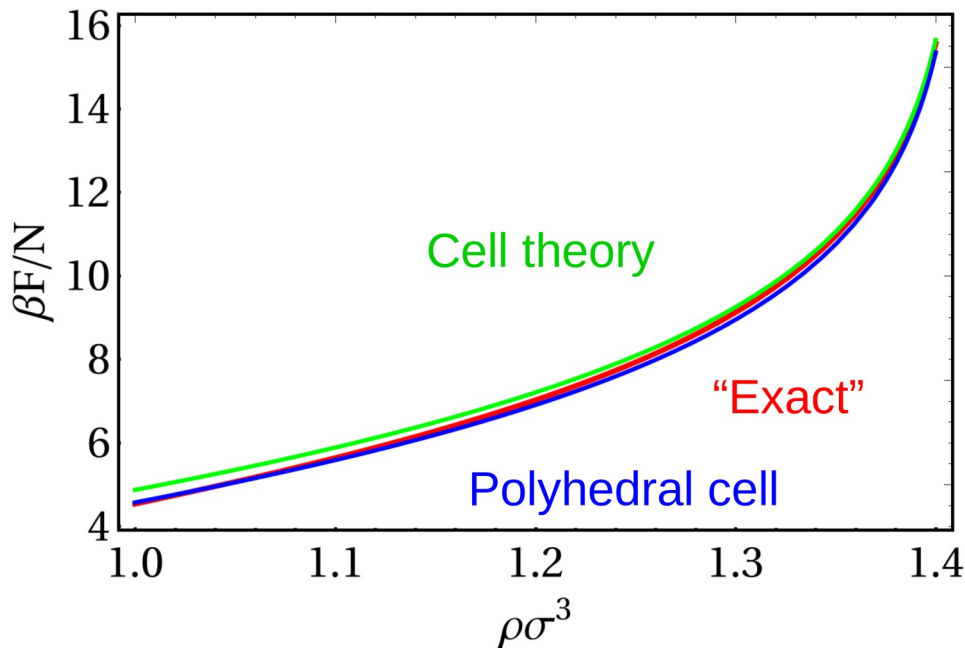
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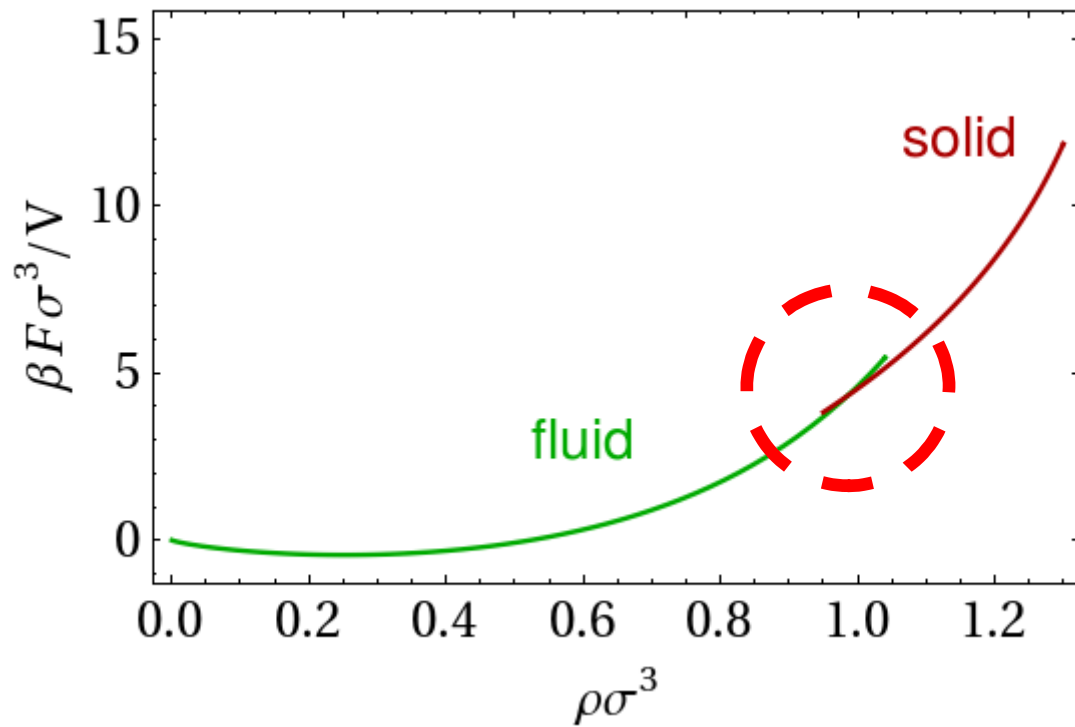
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Full free energy:

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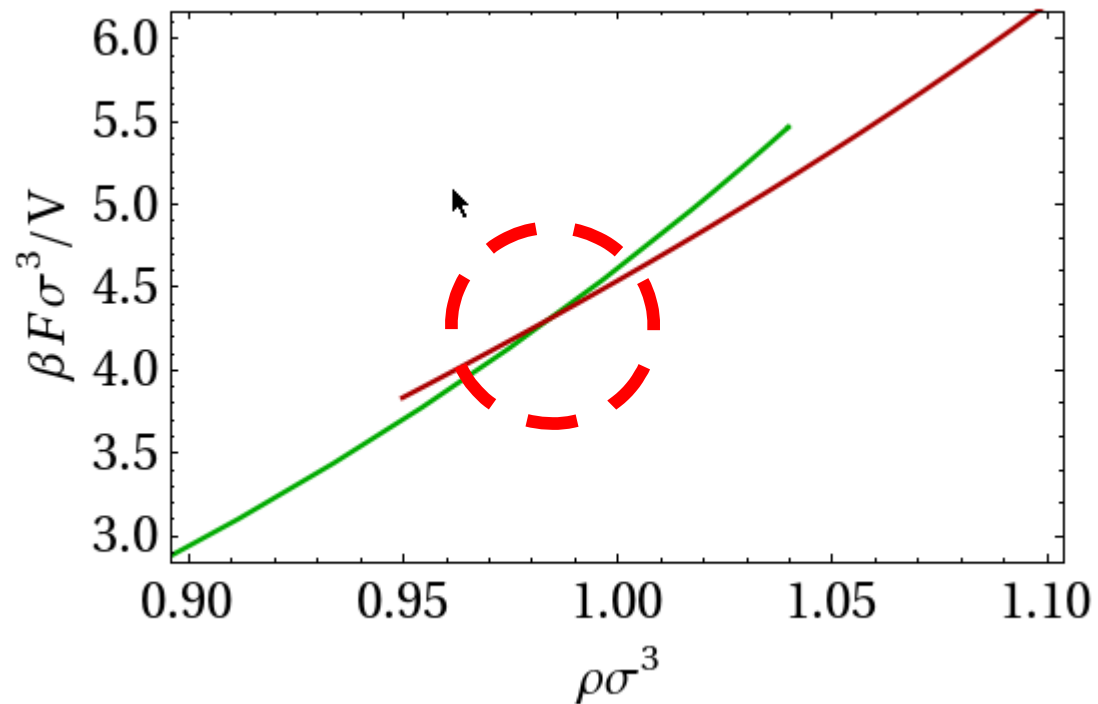
# Phase transition

We can now plot the **free energy per unit volume** of both phases:



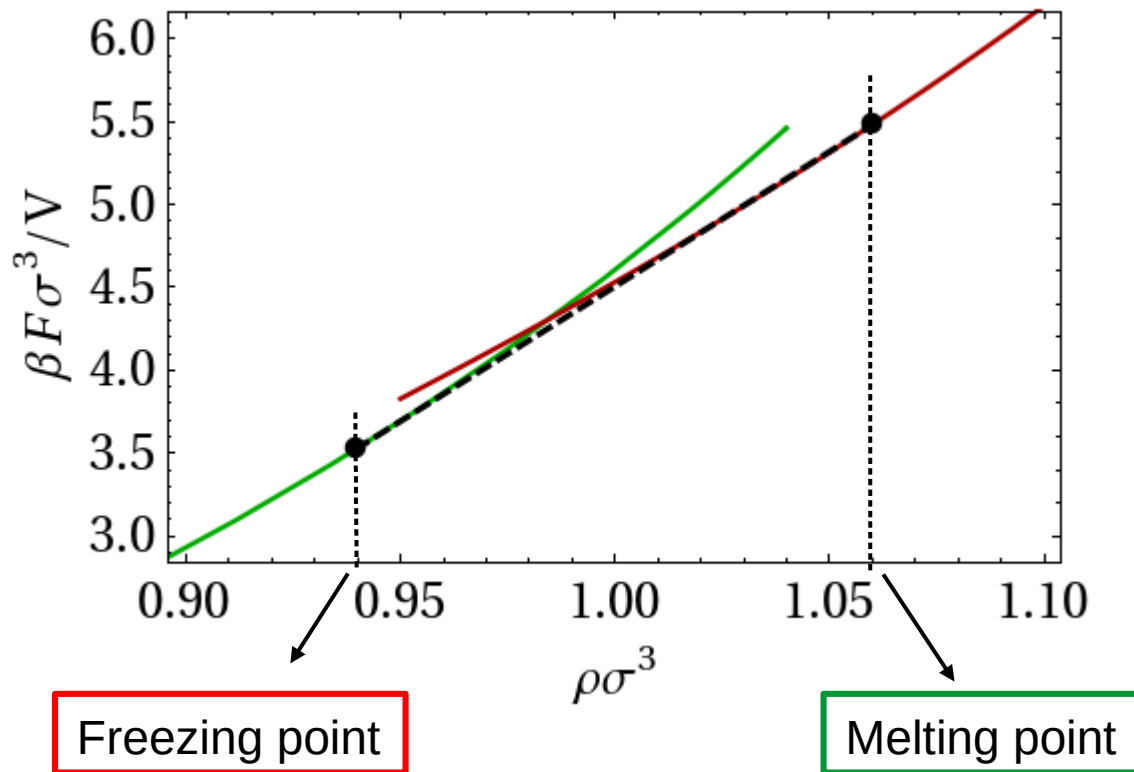
# Phase transition

We can now plot the **free energy per unit volume** of both phases:



# Phase transition

We can now plot the **free energy per unit volume** of both phases:



Coexistence region  
between fluid and solid.

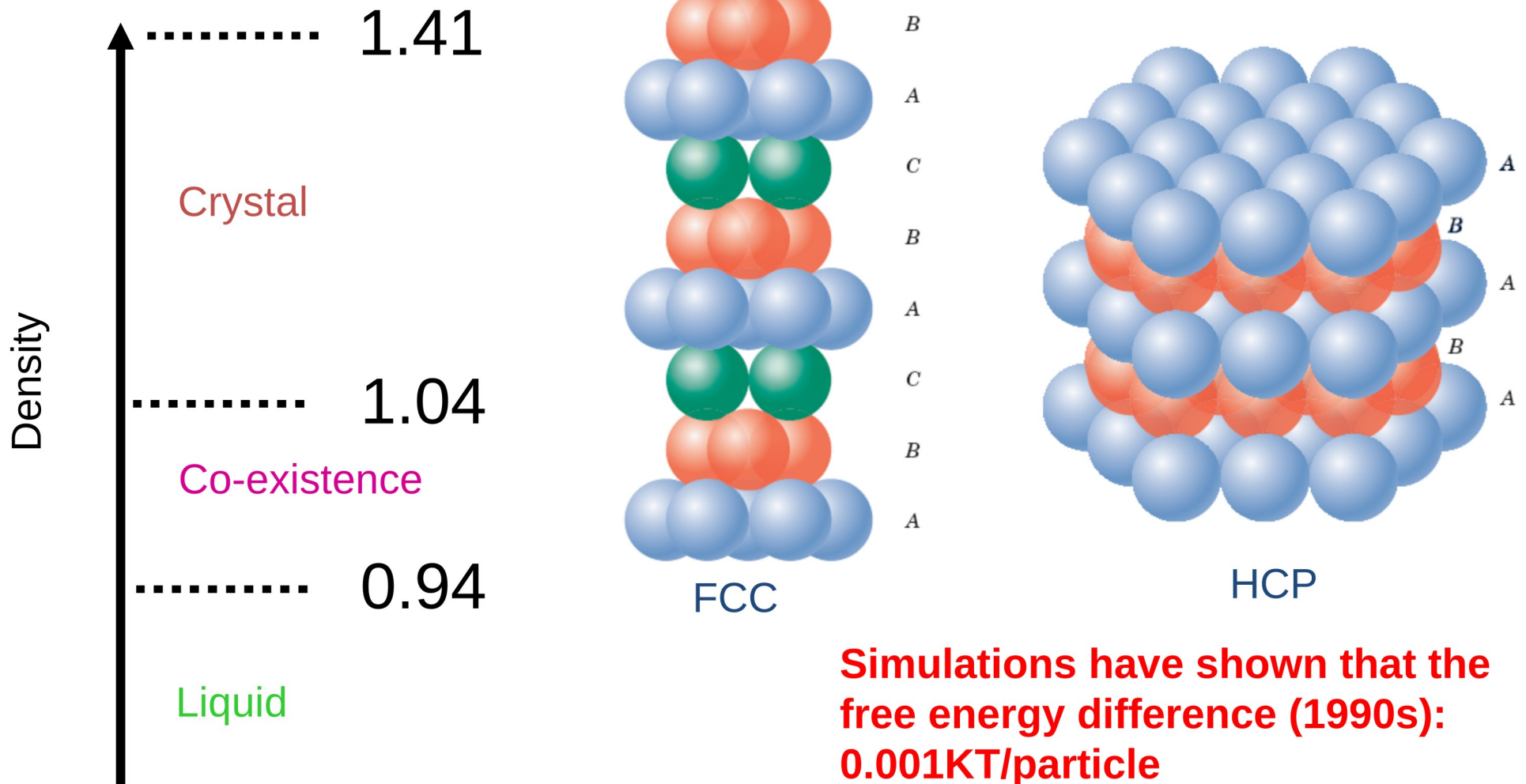
With full simulations:

$$\rho_f \sigma^3 = 0.942, \rho_m \sigma^3 = 1.041$$

With (polyhedral) cell theory:

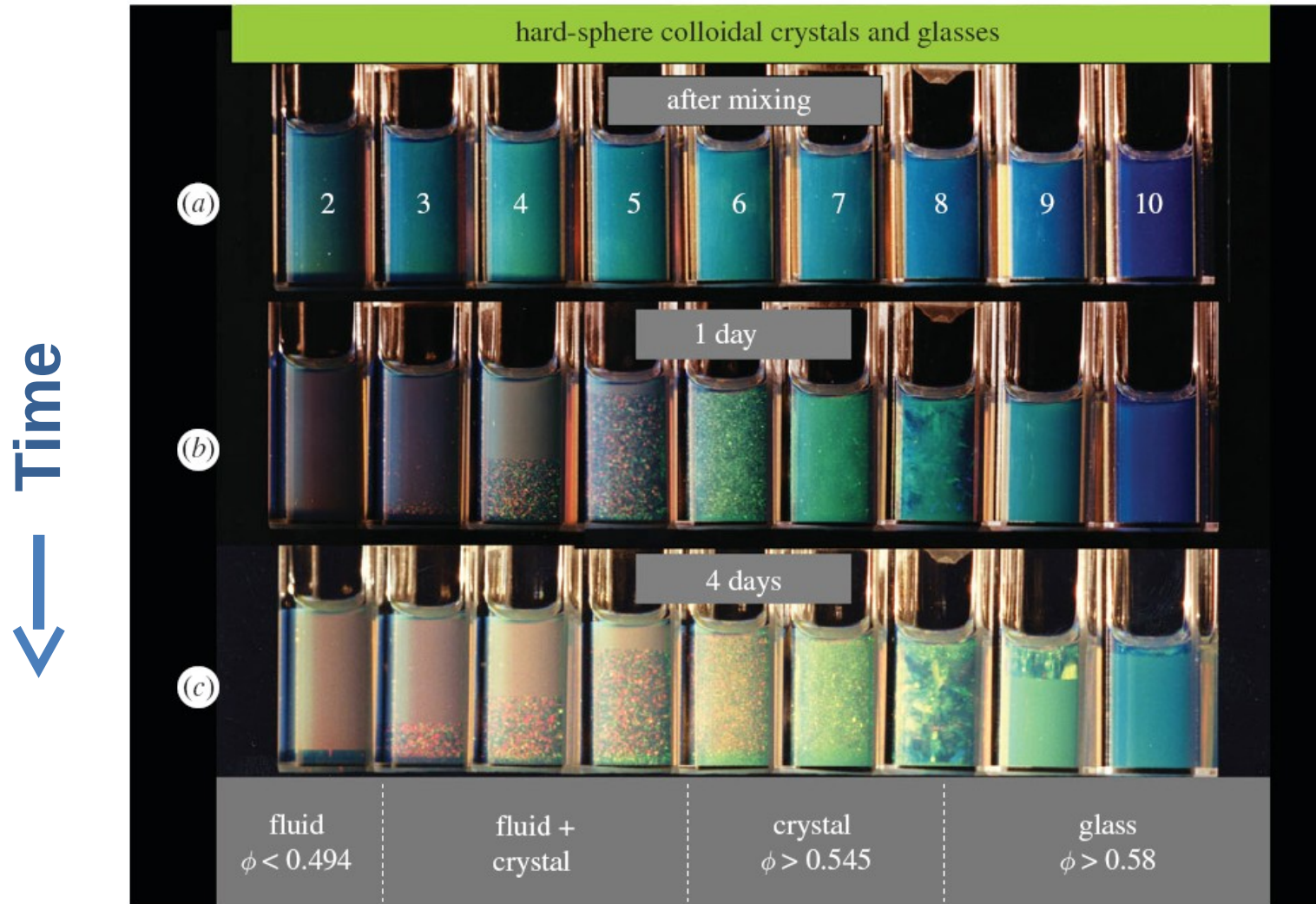
$$\rho_f \sigma^3 = 0.940, \rho_m \sigma^3 = 1.060$$

# Hard sphere phase diagram



**Simulations have shown that the free energy difference (1990s):**  
**0.001KT/particle**  
**FCC wins!**

# Hard sphere phase diagram



Packing fraction →

Figure: Pusey *et al* (2009)  
Remade from 1989



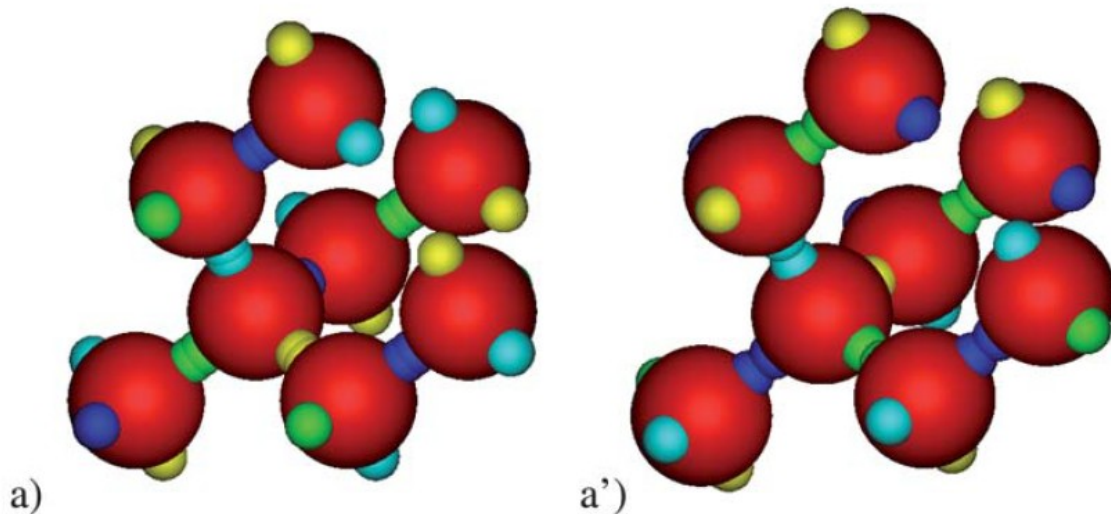
# Free-energy calculation using simulations

---

- 1) Construct a continuous path from a system where you know the free energy to the system where you want to know the free energy.
- 2) Measure free-energy derivative along path
- 3) Calculate free energy

Complications:

- Avoiding phase transitions
- “**Configurational entropy**”

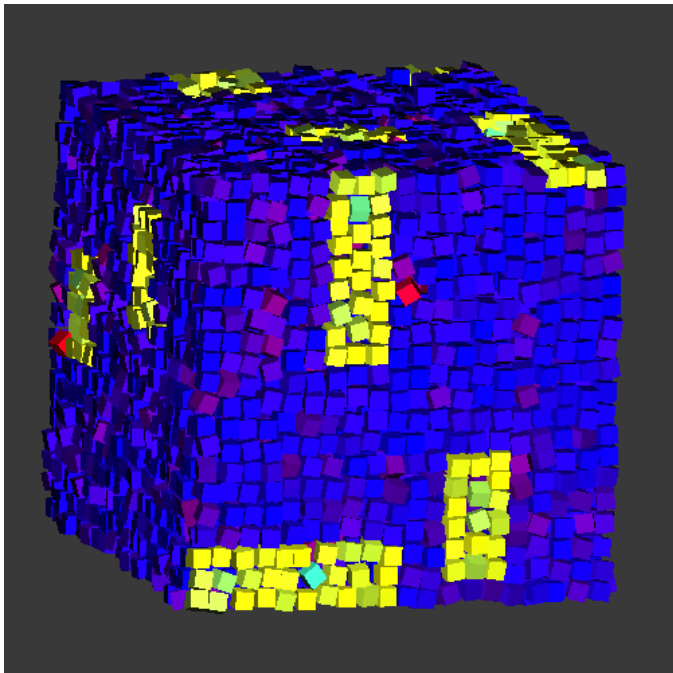
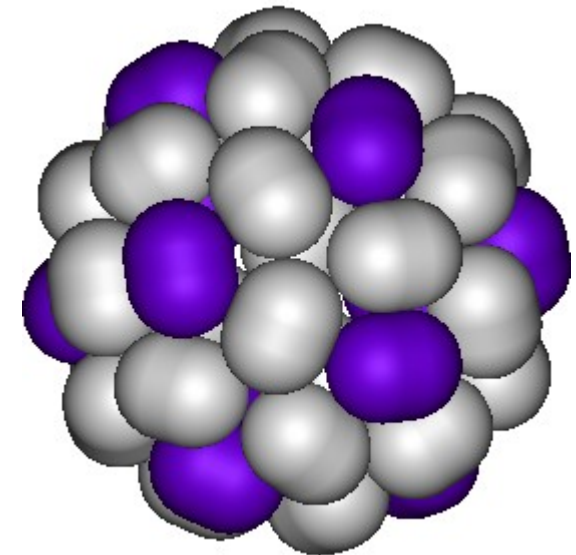
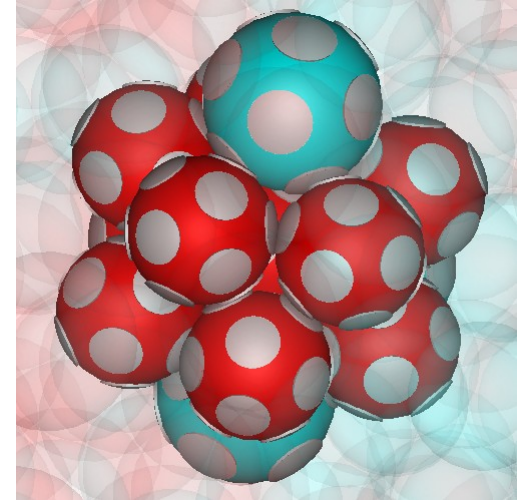


# Adding more length scales

---

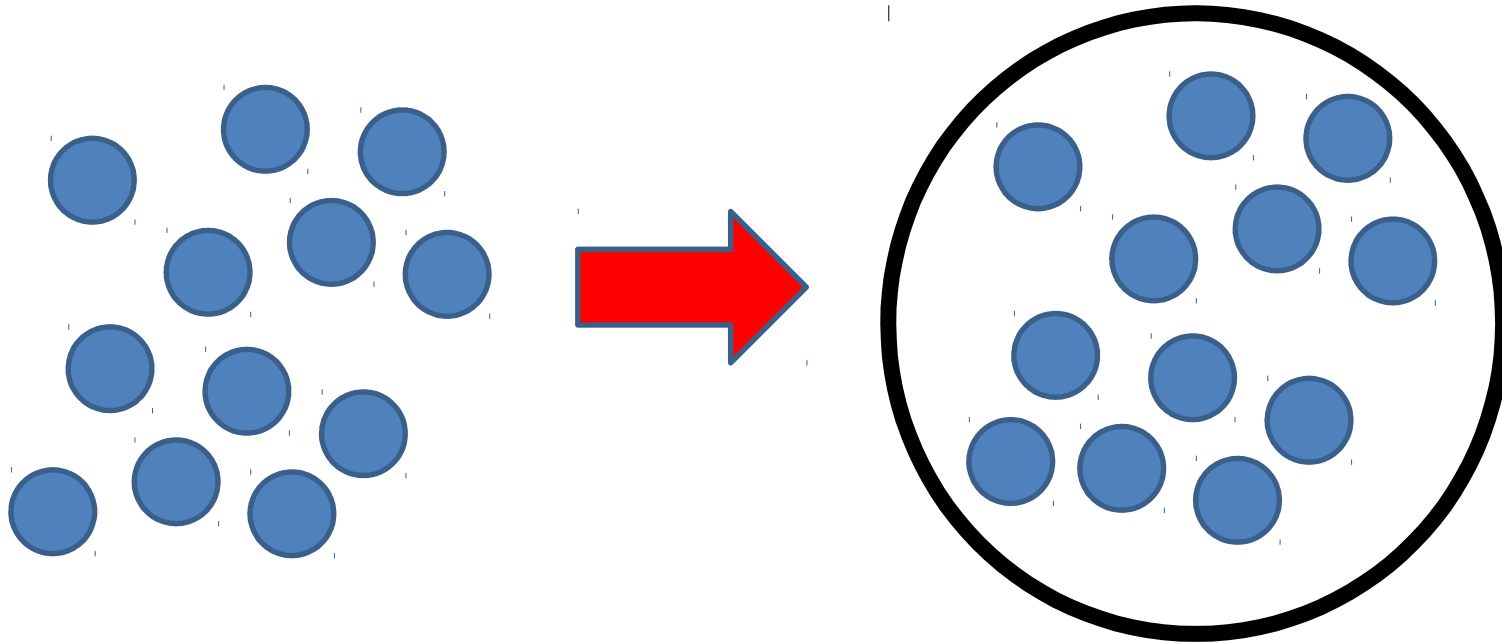
Multiple ways of increasing complexity of the geometry:

- **Confinement**
- **Size mixtures**
- **Soft interactions**
- **Directional interactions**
- ...



# Geometry through confinement

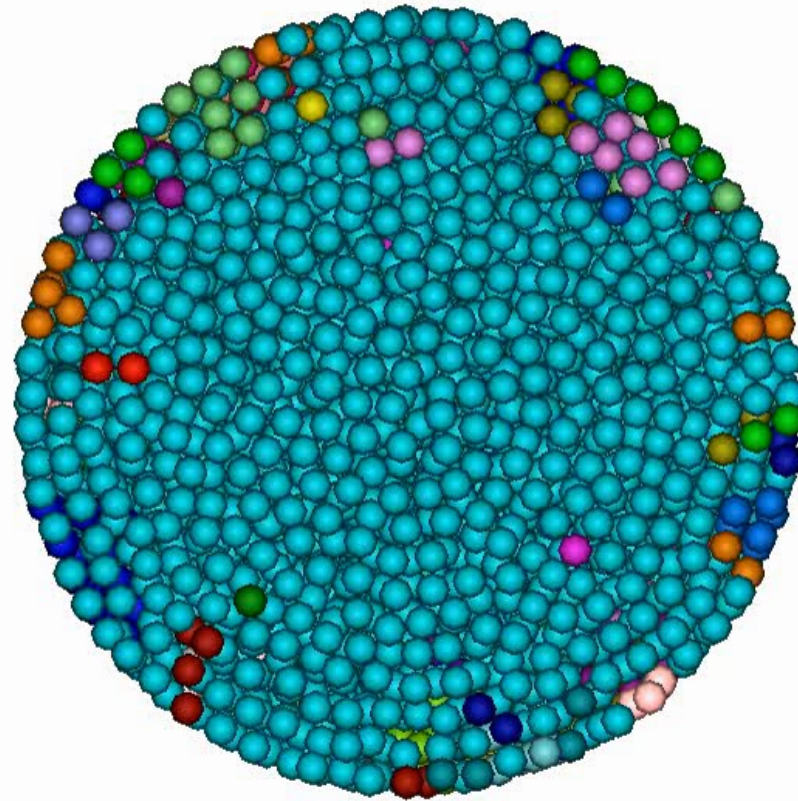
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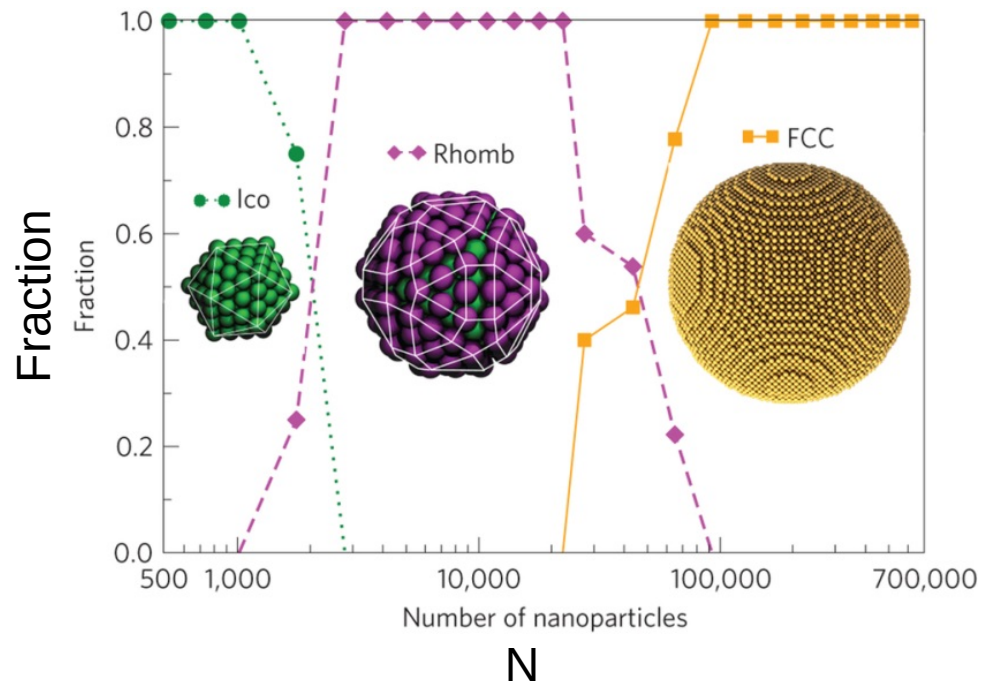
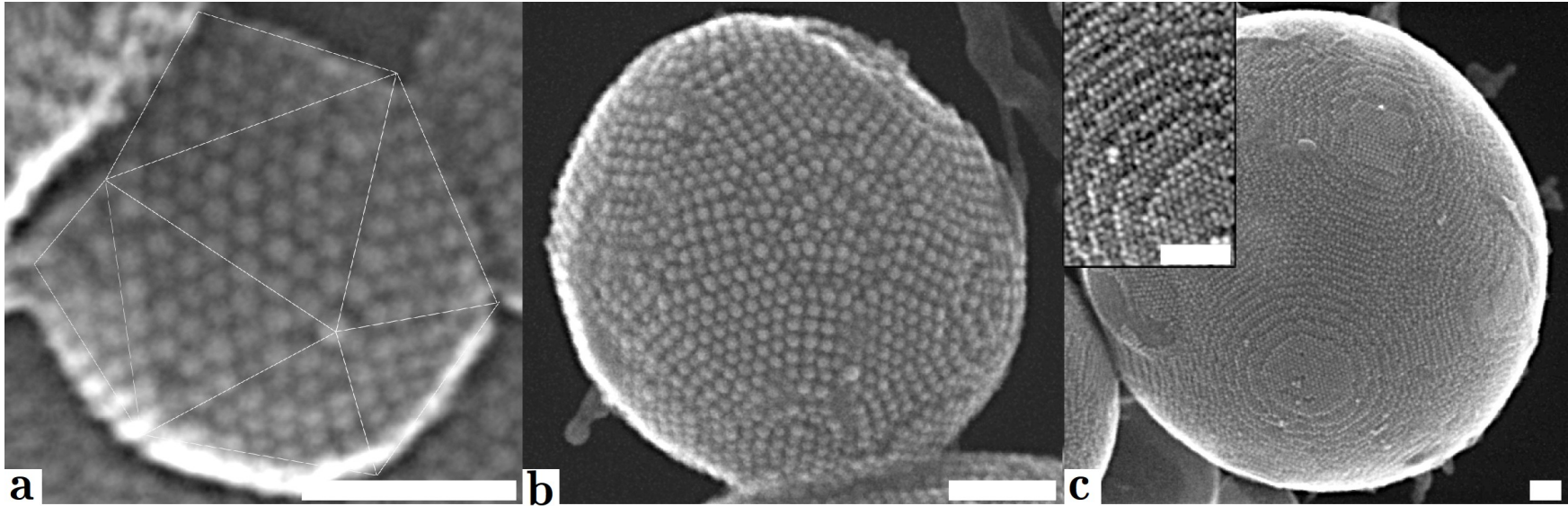
Colloids/nanoparticles confined in shrinking droplets.

# Geometry through confinement

---



# Geometry through confinement

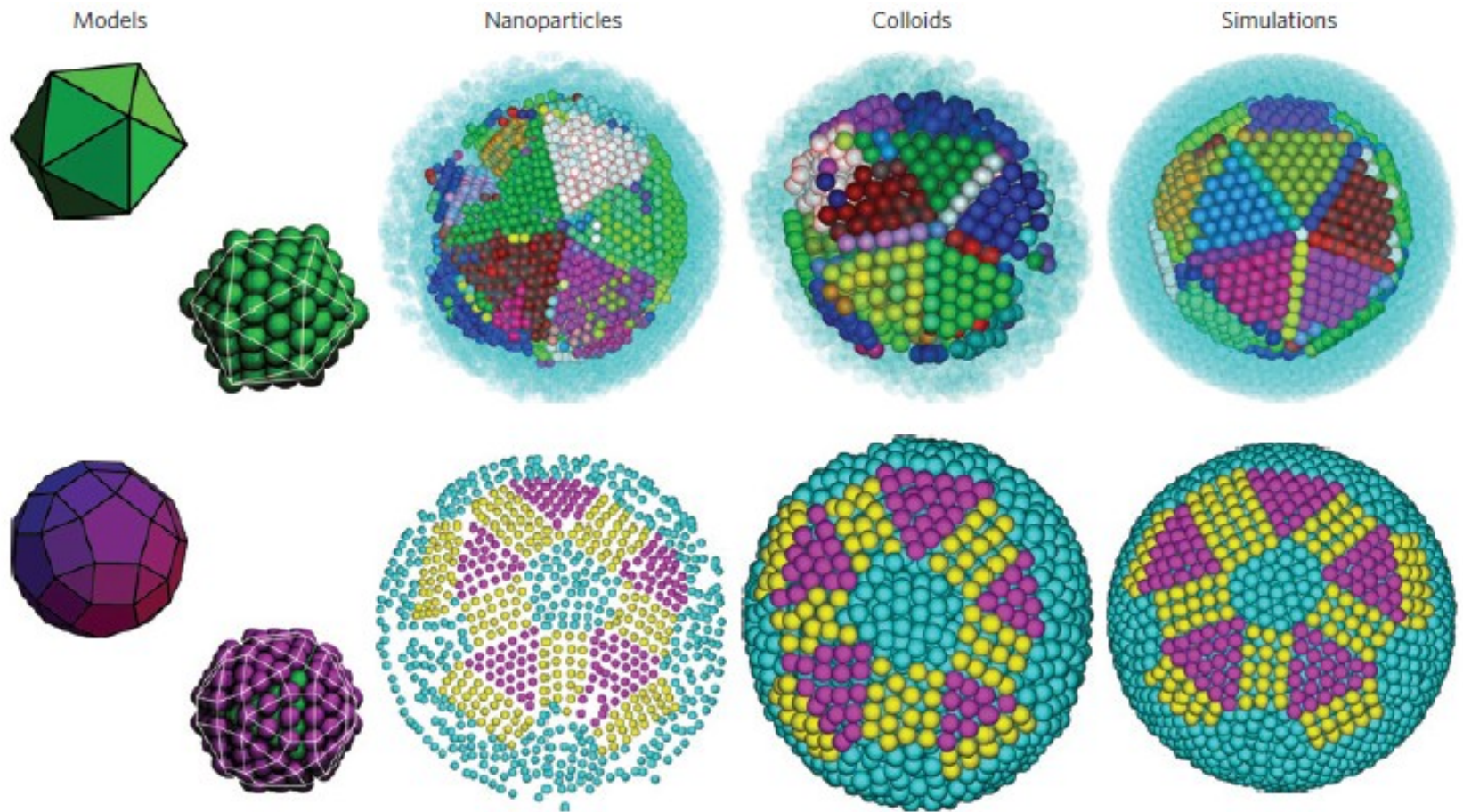


De Nijs et al., *Nature Materials*  
**14**, 56 (2015)

# Geometry through confinement

---

Simulations vs. experiment:



# Candidate crystal structures?

---

We can (in principle) calculate free energies for any relevant crystal structure.

How to determine **what crystal structures are relevant?**

- **Direct simulation (and hope for the best)**
- **Intelligent guessing**
- **Genetic algorithms**
- **Optimization algorithms**

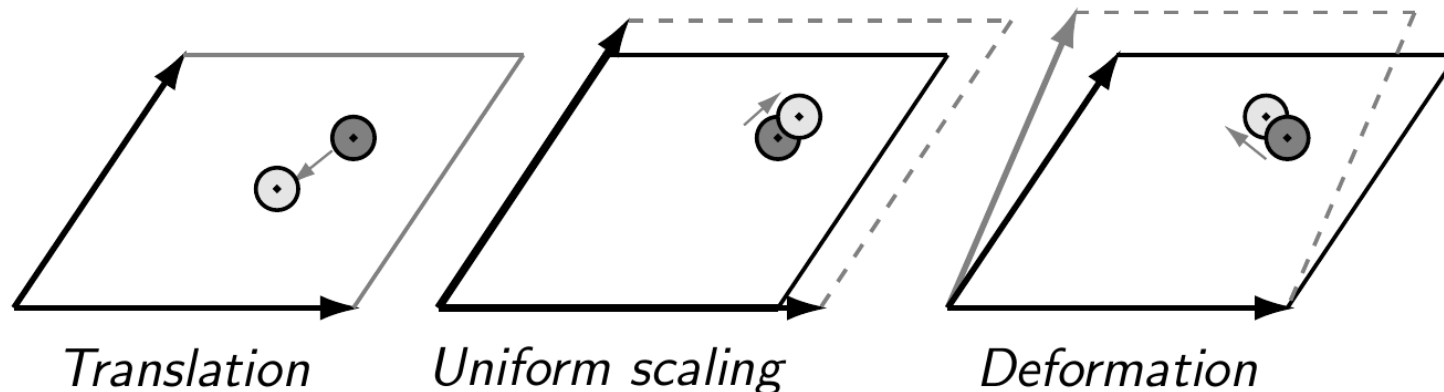
# Candidate crystal structures?

We can (in principle) calculate free energies for any relevant crystal structure.

How to determine **what crystal structures are relevant?**

“Floppy box” Monte Carlo:

- Simulate very few particles in a simulation cell with variable shape.
- Look for structures with high packing, low energy for a wide range of **pressure**, **temperature**, and **number of particles**.

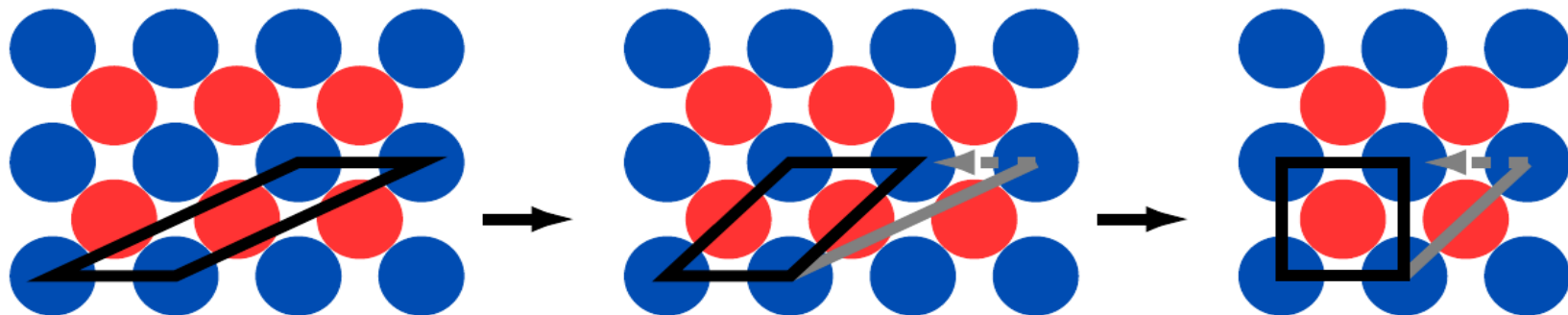




# Rhombohedral unit cell

---

- Resulting “unit cell” can in principle represent any periodic lattice if the number of particles is a multiple of the number of particles in the unit cell.
- No quasicrystals!
- To prevent extremely distorted cells, lattice reduction techniques can be used:



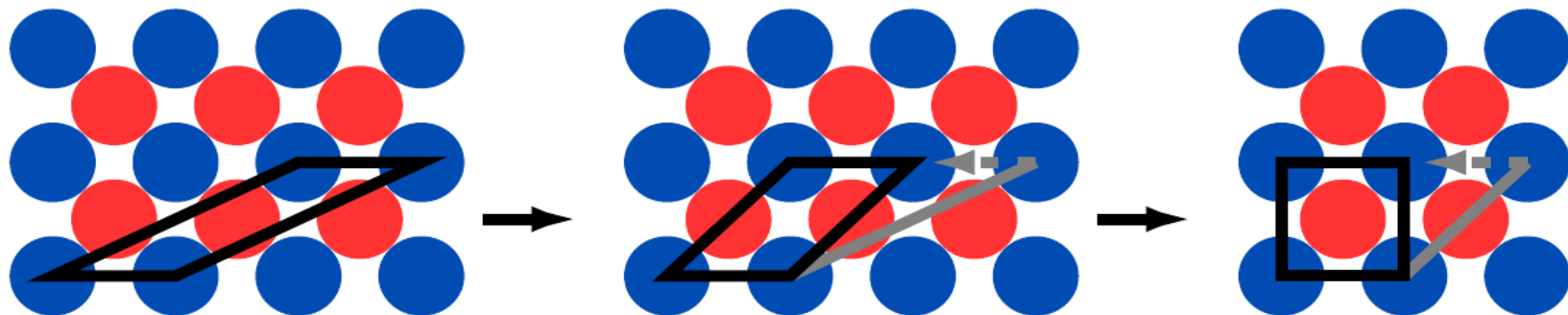
Many separate runs to find possible structures.

Each run is relatively short, and ends with a low-temperature annealing step to suppress thermal fluctuations.

# Rhombohedral unit cell

---

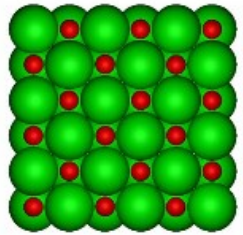
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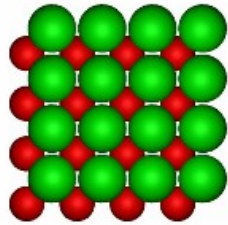
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Each run is relatively short, and ends with a low-temperature annealing step to suppress thermal fluctuations.

# Finding close-packed structures

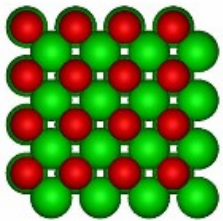


A)

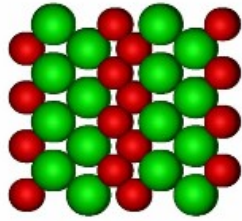


B)

NaCl

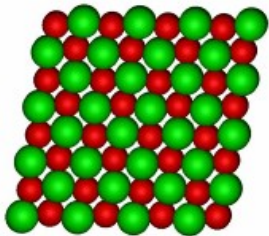


C)

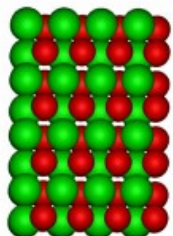


D)

CsCl

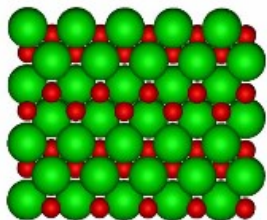


E)

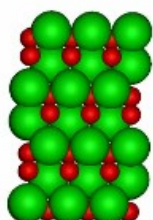


F)

$\gamma$ CuTi



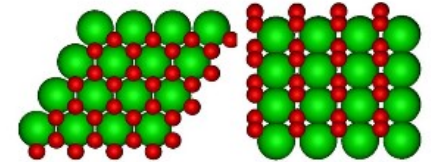
G)



H)

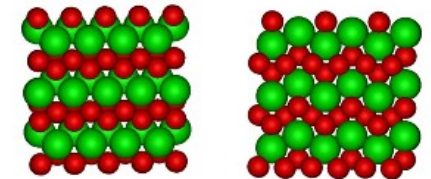
$\alpha$ IrV

AlB<sub>2</sub>

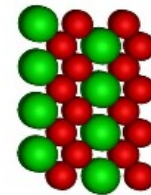


A)

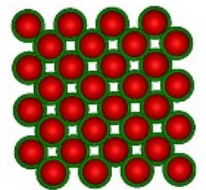
HgBr<sub>2</sub>



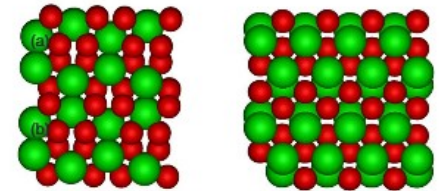
B)



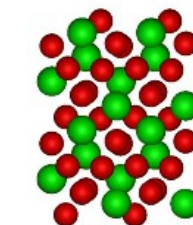
$\alpha\Phi$



AuTe<sub>2</sub>



(a)



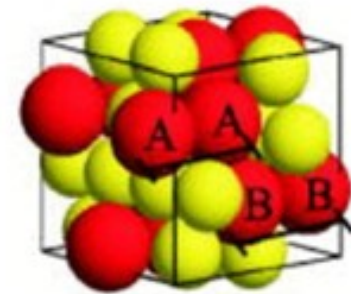
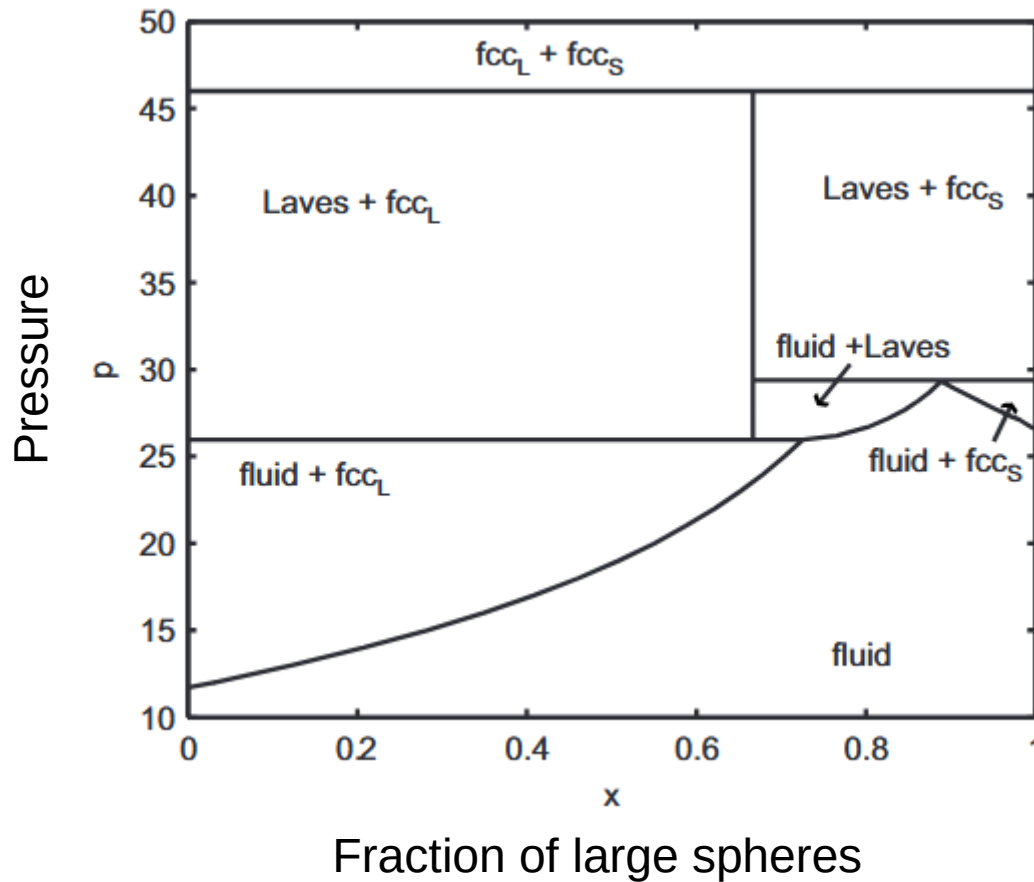
(b)



Filion & Dijkstra, PRE **79**,  
046714 (2009)

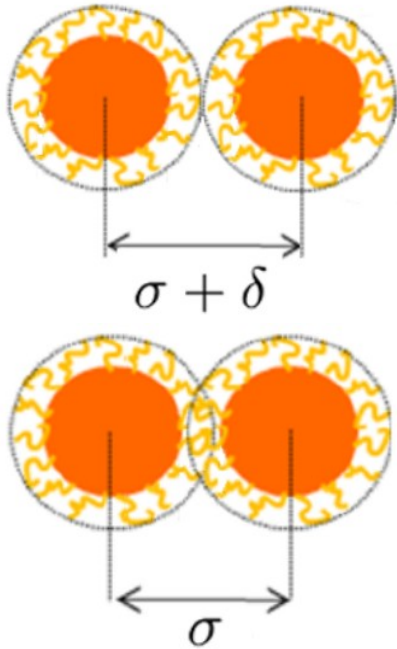
# Not all stable phases are close-packed

Entropy can be higher in phases that do not pack as well.

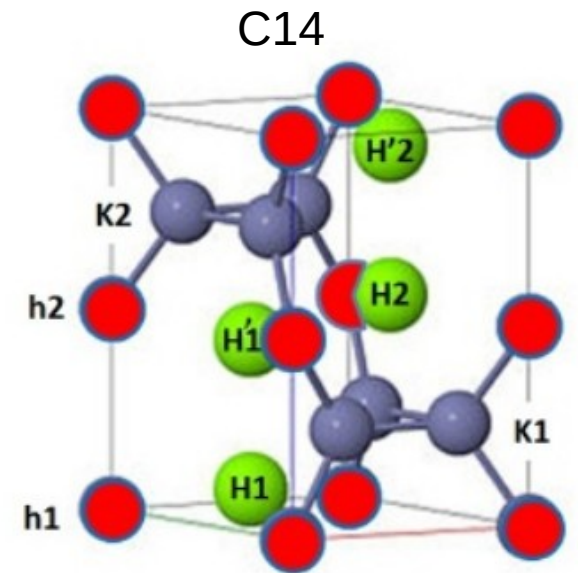
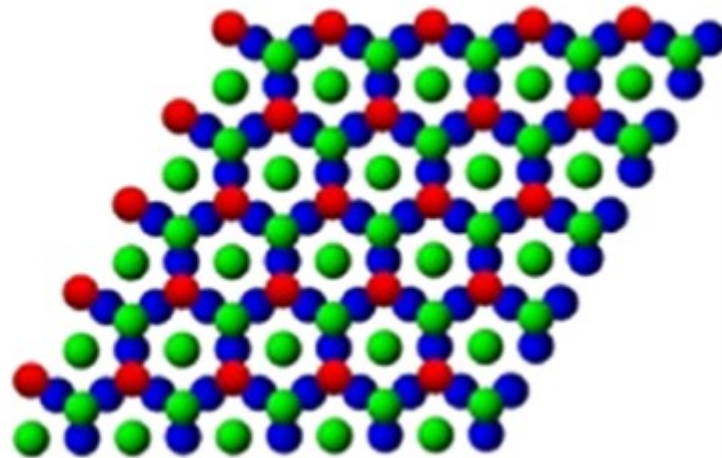
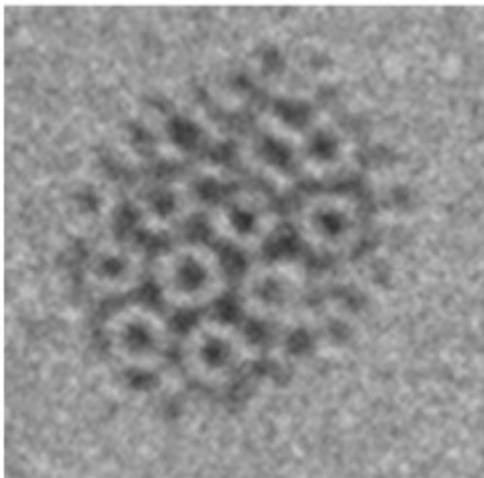


MgZn<sub>2</sub>

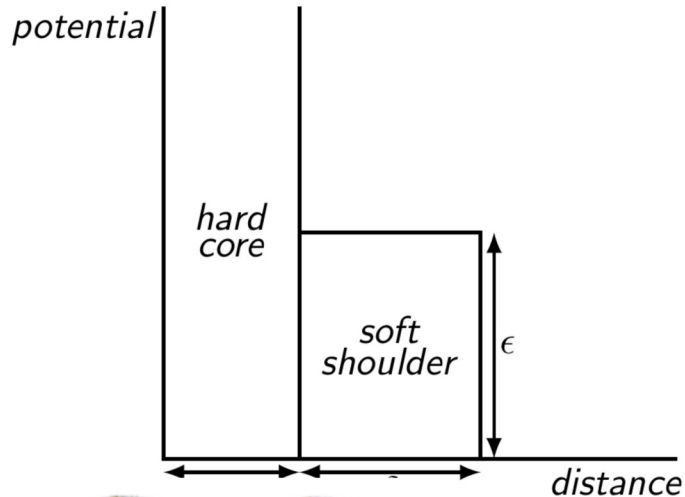
# Geometry through soft interactions



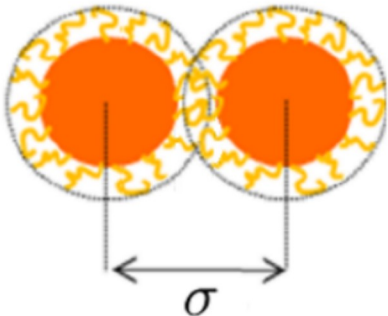
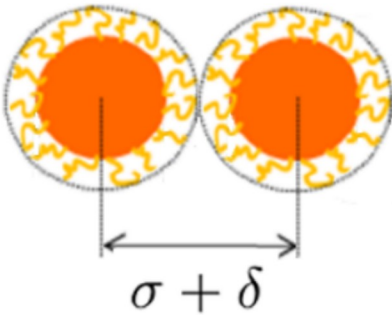
Nanoparticles coated with soft ligands



# Geometry through soft interactions



$$U(r) = \begin{cases} \infty & r < \sigma \\ \epsilon & \sigma < r < \sigma + \delta \\ 0 & r > \sigma + \delta \end{cases}$$



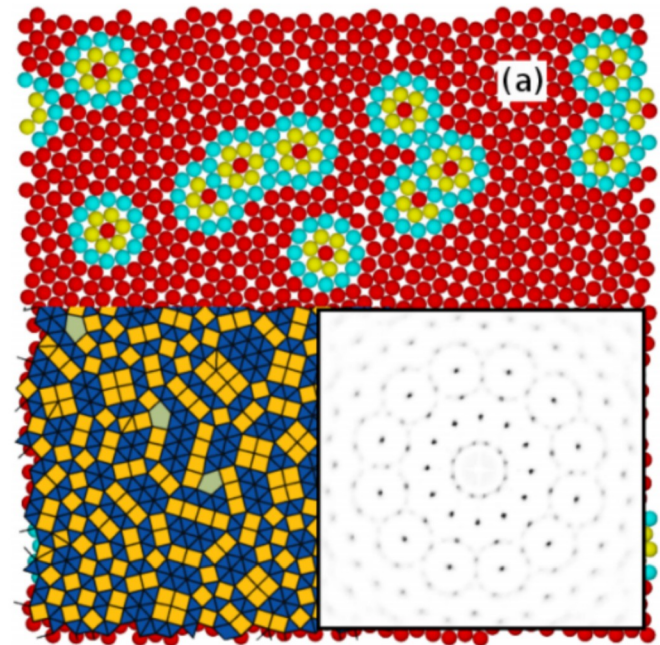
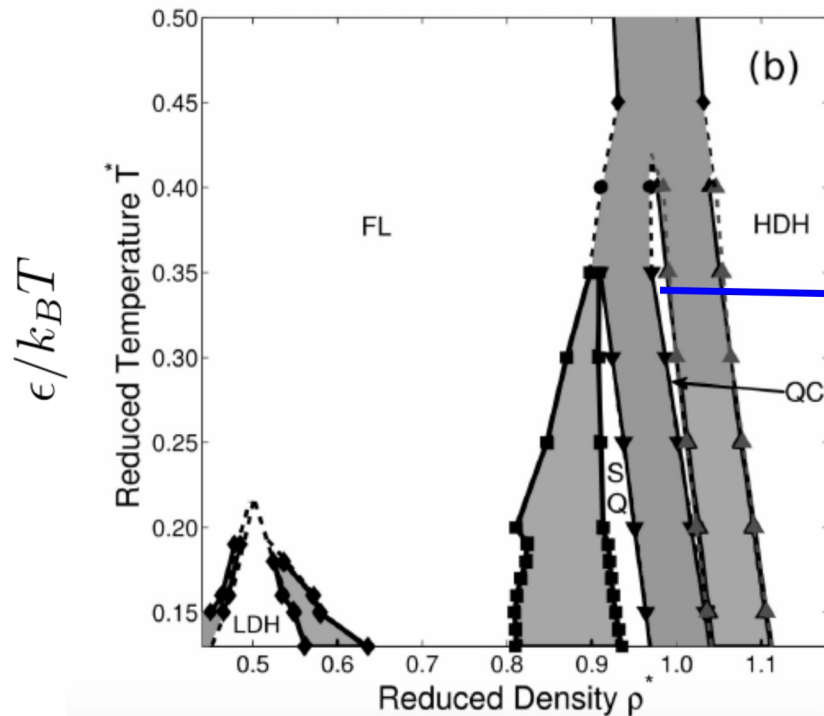
Very rough model for **polymer-coated nanoparticles**:

- fixed energy cost for interpenetration of the polymers
- hard repulsion at some smaller distance  $\sigma$

# Interesting phase behavior?

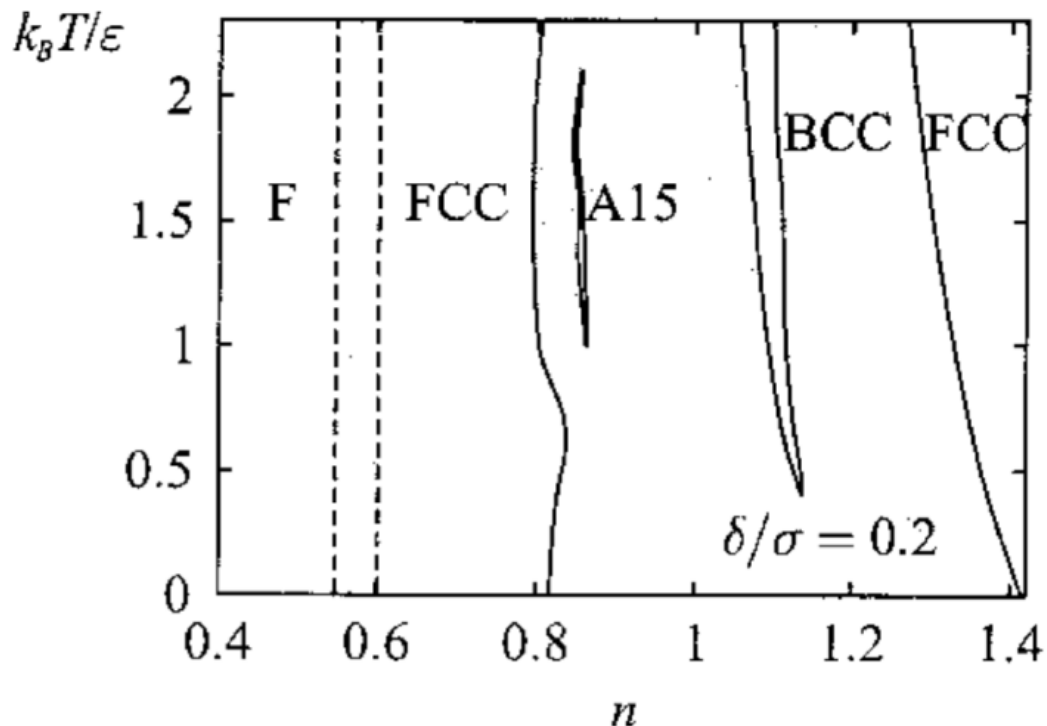
In two dimensions: quasicrystals!

$$\delta/\sigma = 0.4$$



# Interesting phase behavior?

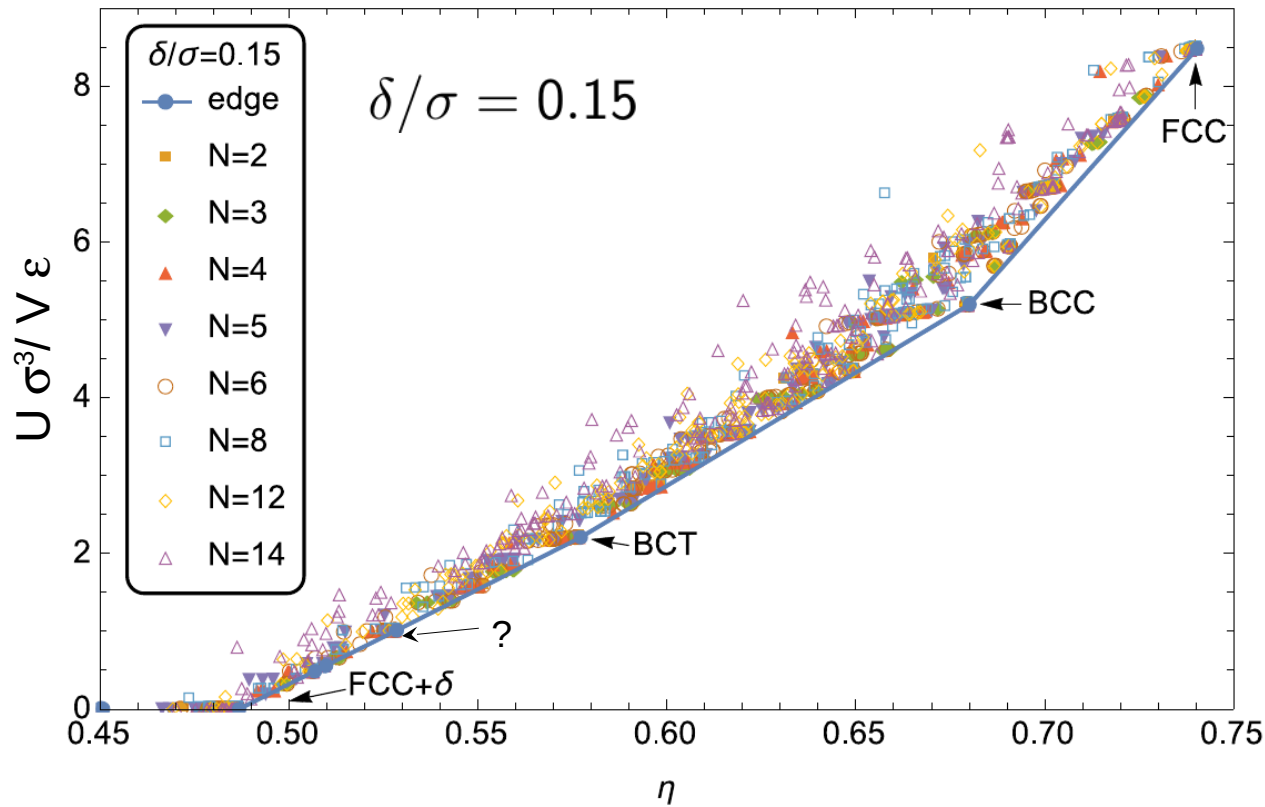
In three dimensions: complex crystals?



- Phase diagram based on a (different) mean-field approach
- Only considered a few different crystal structures
- Effect of temperature on fluid stability neglected



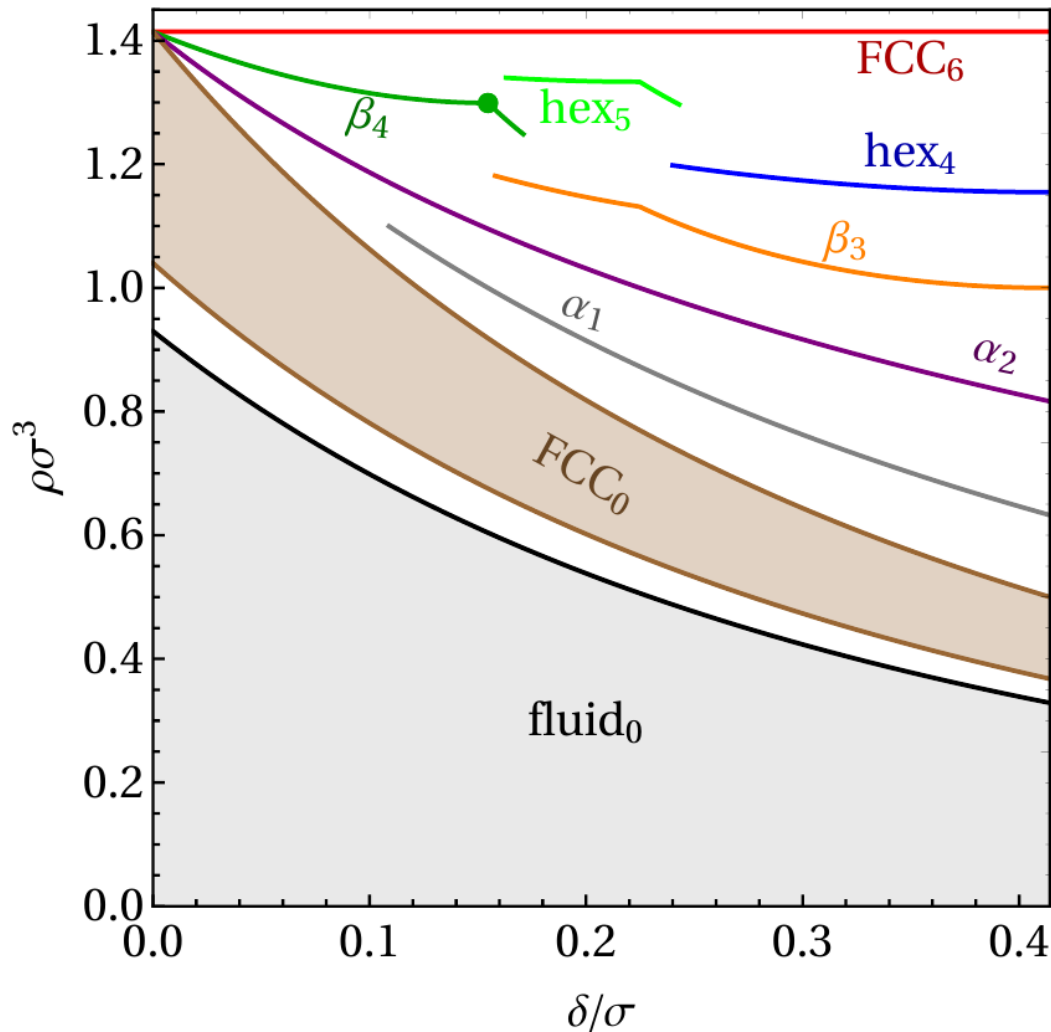
# Candidate structures



Sample output:

- Each point is an MC run
- Bottom envelope is the lowest-energy state (or coexistence) possible at each density
- Several structures clearly stable at zero-temperature.

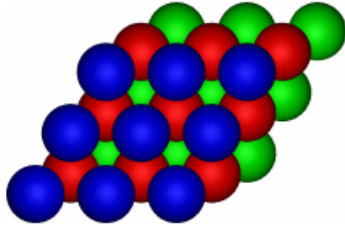
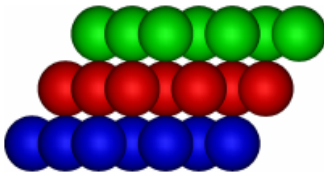
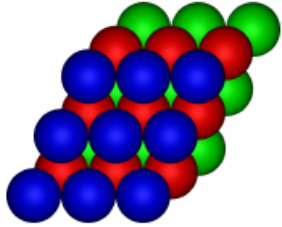
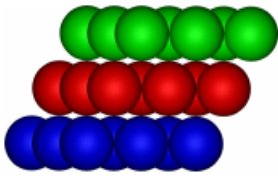
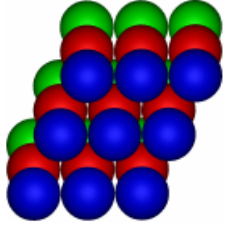
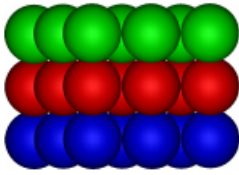
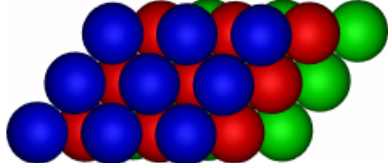
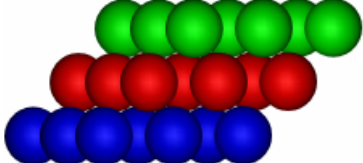
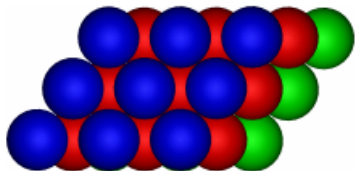
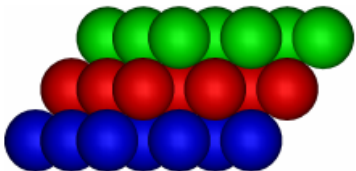
# Zero-T structures



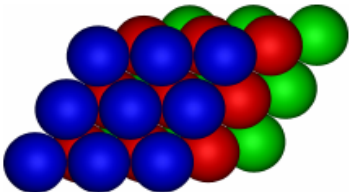
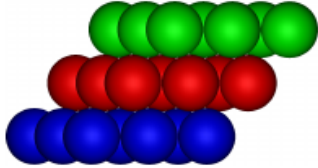
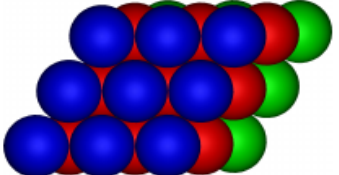
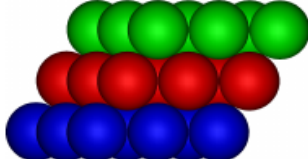
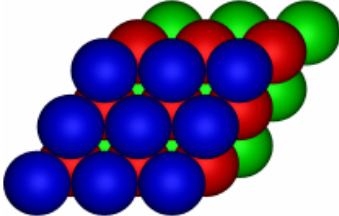
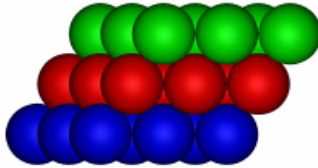
For each shoulder length, we map out all structures found to be stable in the zero-T limit.

- No Frank-Kasper phases found to be stable
- Large variety of structures, but all with simple unit cells
- Subscript indicates energy per particle

# Zero-T structures

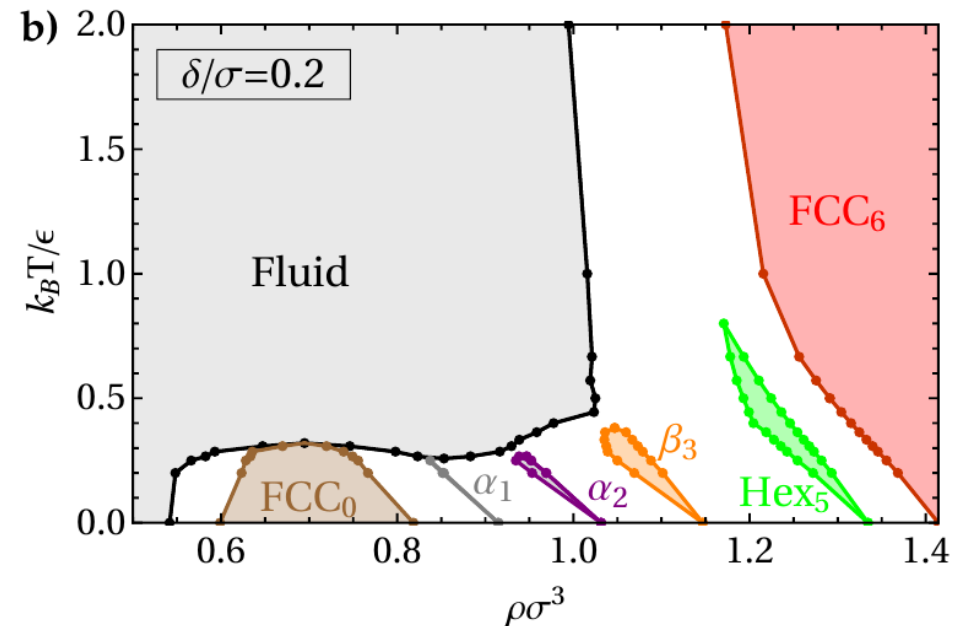
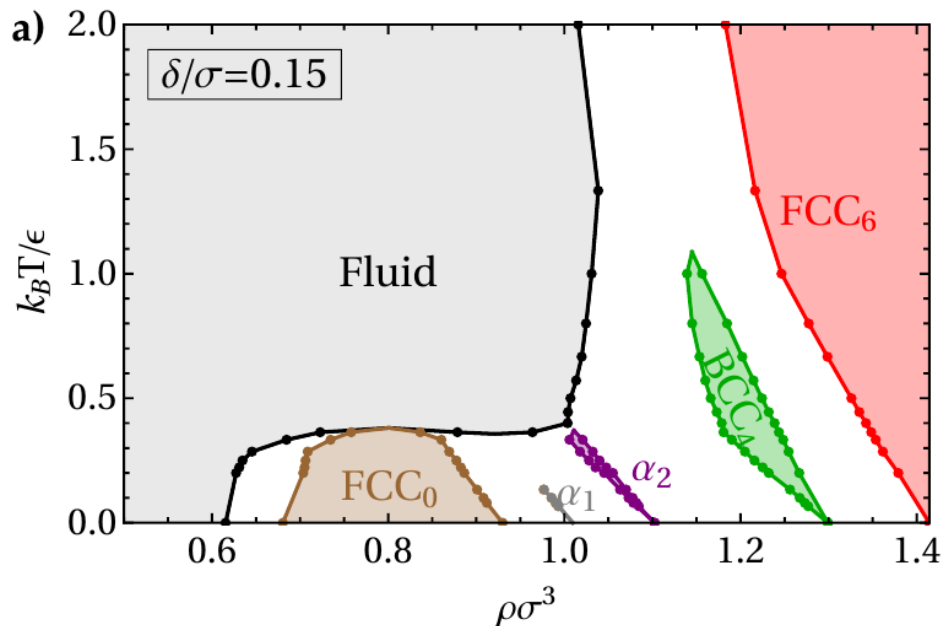
Name	Front view	Side view	$U/N\epsilon$	Notes
$FCC_0$			0	Low-density FCC.
$\alpha_1$			1	Body-centered orthogonal (BCO) lattice. Also contains low-density hexagonal planes.
$\alpha_2$			2	Body-centered tetragonal lattice. Also contains square and $\beta$ planes.
$\beta_3$			3	Also contains $\alpha$ planes.
$\beta_4$			4	Body-centered tetragonal (BCT) lattice. Body-centered cubic when $\delta = \frac{2}{\sqrt{3}} - 1 \simeq 0.1547$ .

# Zero-T structures

Name	Front view	Side view	$U/N\epsilon$	Notes
hex <sub>4</sub>			4	Also contains $\beta$ planes.
hex <sub>5</sub>			5	Body-centered orthogonal (BCO) lattice.
FCC <sub>6</sub>			6	High-density FCC.

# Extension to finite temperature

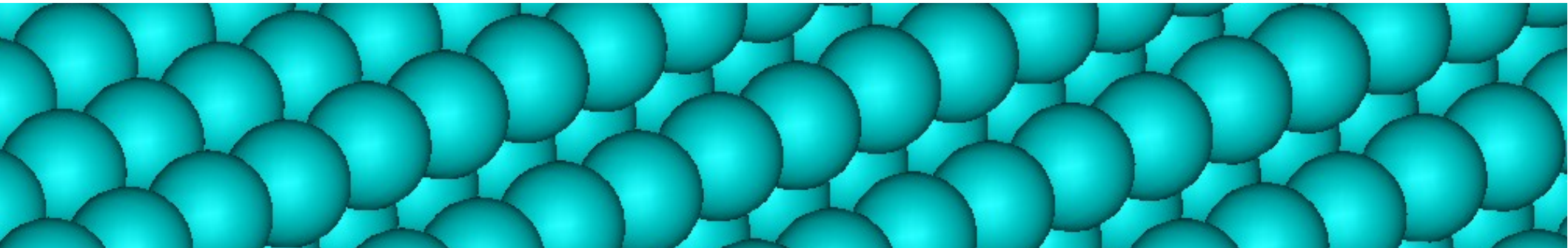
Using **cell theory**, we can draw the finite-temperature phase diagram for specific shoulder lengths.



# Conclusions

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- Strong connection between geometry and entropy
- Even simple interactions can lead to complex structures
- Resulting structures can be hard to guess!





# Geometry through size mixtures

